

2-Nonene, 4,6-dimethyl, # 2

| | |
|----------------------|--|
| Inchi: | InChI=1S/C11H22/c1-5-7-10(3)9-11(4)8-6-2/h5,7,10-11H,6,8-9H2,1-4H3/b7-5+ |
| InchiKey: | ONGPAIHWJMXLYLD-FNORWQNLSA-N |
| Formula: | C11H22 |
| SMILES: | CC=CC(C)CC(C)CCC |
| Mol. weight [g/mol]: | 154.29 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 117.08 | kJ/mol | Joback Method |
| hf | -163.71 | kJ/mol | Joback Method |
| hfus | 17.40 | kJ/mol | Joback Method |
| hvap | 39.26 | kJ/mol | Joback Method |
| log10ws | -3.80 | | Crippen Method |
| logp | 4.025 | | Crippen Method |
| mcvol | 161.550 | ml/mol | McGowan Method |
| pc | 2054.89 | kPa | Joback Method |
| rinpola | 1000.00 | | NIST Webbook |
| tb | 454.36 | K | Joback Method |
| tc | 630.87 | K | Joback Method |
| tf | 178.65 | K | Joback Method |
| vc | 0.620 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 344.70 | J/molxK | 454.36 | Joback Method |
| cpg | 420.97 | J/molxK | 601.45 | Joback Method |
| cpg | 407.08 | J/molxK | 572.03 | Joback Method |
| cpg | 392.54 | J/molxK | 542.62 | Joback Method |
| cpg | 377.31 | J/molxK | 513.20 | Joback Method |
| cpg | 361.37 | J/molxK | 483.78 | Joback Method |
| cpg | 434.23 | J/molxK | 630.87 | Joback Method |
| dvisc | 0.0001718 | Paxs | 454.36 | Joback Method |
| dvisc | 0.0002429 | Paxs | 408.41 | Joback Method |

| | | | | |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0003749 | Paxs | 362.46 | Joback Method |
| dvisc | 0.0006563 | Paxs | 316.50 | Joback Method |
| dvisc | 0.0013897 | Paxs | 270.55 | Joback Method |
| dvisc | 0.0040002 | Paxs | 224.60 | Joback Method |
| dvisc | 0.0198348 | Paxs | 178.65 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=R568324&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

Latest version available from:

<https://www.chemeo.com/cid/53-258-7/2-Nonene-4-6-dimethyl-2.pdf>

Generated by Cheméo on 2024-04-25 07:04:13.538526728 +0000 UTC m=+16317902.459104040.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.