

# Benzene, 1,2-dimethoxy-4-nitro-

<b>Other names:</b>	3,4-Dimethoxynitrobenzene 1,2-Dimethoxy-4-nitrobenzene 4-Nitroveratrole 4-Nitroveratrole 2-Methoxy-5-nitrophenol, methyl ether
<b>Inchi:</b>	InChI=1S/C8H9NO4/c1-12-7-4-3-6(9(10)11)5-8(7)13-2/h3-5H,1-2H3
<b>InchiKey:</b>	YFWBUVZWCBFSQN-UHFFFAOYSA-N
<b>Formula:</b>	C8H9NO4
<b>SMILES:</b>	COc1ccc([N+](=O)[O-])cc1OC
<b>Mol. weight [g/mol]:</b>	183.16
<b>CAS:</b>	709-09-1

## Physical Properties

Property code	Value	Unit	Source
gf	-64.82	kJ/mol	Joback Method
hf	-270.06	kJ/mol	Joback Method
hfus	23.48	kJ/mol	Joback Method
hvap	58.41	kJ/mol	Joback Method
log10ws	-2.36		Crippen Method
logp	1.612		Crippen Method
mcvol	128.980	ml/mol	McGowan Method
pc	3431.89	kPa	Joback Method
rinpol	1624.10		NIST Webbook
tb	615.76	K	Joback Method
tc	854.02	K	Joback Method
tf	419.45	K	Joback Method
vc	0.493	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	309.07	J/mol×K	615.76	Joback Method
cpg	320.70	J/mol×K	655.47	Joback Method
cpg	331.62	J/mol×K	695.18	Joback Method

cpg	341.82	J/mol×K	734.89	Joback Method
cpg	351.29	J/mol×K	774.60	Joback Method
cpg	360.02	J/mol×K	814.31	Joback Method
cpg	368.00	J/mol×K	854.02	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	503.20	K	2.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C709091&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C709091&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/53-279-4/Benzene-1-2-dimethoxy-4-nitro.pdf>

Generated by Cheméo on 2024-04-27 07:22:20.073477001 +0000 UTC m=+16491788.994054327.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.