

# [2,3-2H2]-trans-2,3-Epoxyoctanal

Inchi:	InChI=1S/C8H14O2/c1-2-3-4-5-7-8(6-9)10-7/h6-8H,2-5H2,1H3/t7-,8-/m1/s1/i7D,8D
InchiKey:	YWFUECKBUFORTA-YAZXJRJDSA-N
Formula:	C8H12D2O2
SMILES:	CCCCC1OC1C=O
Mol. weight [g/mol]:	144.21

## Physical Properties

Property code	Value	Unit	Source
gf	-116.12	kJ/mol	Joback Method
hf	-373.57	kJ/mol	Joback Method
hfus	25.95	kJ/mol	Joback Method
hvap	44.24	kJ/mol	Joback Method
log10ws	-1.66		Crippen Method
logp	1.533		Crippen Method
mcvol	120.160	ml/mol	McGowan Method
pc	3005.73	kPa	Joback Method
ripol	1093.00		NIST Webbook
ripol	1533.00		NIST Webbook
ripol	1533.00		NIST Webbook
ripol	1542.00		NIST Webbook
tb	460.12	K	Joback Method
tc	644.46	K	Joback Method
tf	262.19	K	Joback Method
vc	0.477	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	272.10	J/molxK	460.12	Joback Method
cpg	285.30	J/molxK	490.84	Joback Method
cpg	297.85	J/molxK	521.57	Joback Method
cpg	309.78	J/molxK	552.29	Joback Method
cpg	321.11	J/molxK	583.02	Joback Method
cpg	331.87	J/molxK	613.74	Joback Method

cpg	342.08	J/molxK	644.46	Joback Method
dvisc	0.0020728	Paxs	262.19	Joback Method
dvisc	0.0015252	Paxs	295.18	Joback Method
dvisc	0.0011937	Paxs	328.17	Joback Method
dvisc	0.0009770	Paxs	361.15	Joback Method
dvisc	0.0008269	Paxs	394.14	Joback Method
dvisc	0.0007181	Paxs	427.13	Joback Method
dvisc	0.0006364	Paxs	460.12	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R396080&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R396080&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>ripolar:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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