

2,6-Dimethoxyphenol, pentafluoropropionate

Inchi: InChI=1S/C11H9F5O4/c1-18-6-4-3-5-7(19-2)8(6)20-9(17)10(12,13)11(14,15)16/h3-5H,1-
InchiKey: IWWKATRLWJSPNZ-UHFFFAOYSA-N
Formula: C11H9F5O4
SMILES: COc1cccc(OC)c1OC(=O)C(F)(F)C(F)(F)F
Mol. weight [g/mol]: 300.18

Physical Properties

Property code	Value	Unit	Source
gf	-1277.40	kJ/mol	Joback Method
hf	-1564.07	kJ/mol	Joback Method
hfus	23.24	kJ/mol	Joback Method
hvap	50.98	kJ/mol	Joback Method
log10ws	-3.42		Crippen Method
logp	2.807		Crippen Method
mcvol	170.120	ml/mol	McGowan Method
pc	2167.36	kPa	Joback Method
rinpol	1304.00		NIST Webbook
tb	598.74	K	Joback Method
tc	783.55	K	Joback Method
tf	389.60	K	Joback Method
vc	0.671	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	442.65	J/mol×K	598.74	Joback Method
cpg	454.35	J/mol×K	629.54	Joback Method
cpg	465.35	J/mol×K	660.34	Joback Method
cpg	475.67	J/mol×K	691.15	Joback Method
cpg	485.32	J/mol×K	721.95	Joback Method
cpg	494.33	J/mol×K	752.75	Joback Method
cpg	502.70	J/mol×K	783.55	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U374279&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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