

# PHENOL, 2,6-DIBROMO-, ACETATE

|                      |                                                                   |
|----------------------|-------------------------------------------------------------------|
| Other names:         | 2,6-Dibromophenyl acetate<br>Acetic acid, 2,6-dibromophenyl ester |
| Inchi:               | InChI=1S/C8H6Br2O2/c1-5(11)12-8-6(9)3-2-4-7(8)10/h2-4H,1H3        |
| InchiKey:            | PMNMDWCAAWUBOG-UHFFFAOYSA-N                                       |
| Formula:             | C8H6Br2O2                                                         |
| SMILES:              | CC(=O)Oc1c(Br)cccc1Br                                             |
| Mol. weight [g/mol]: | 293.94                                                            |

## Physical Properties

| Property code | Value       | Unit   | Source             |
|---------------|-------------|--------|--------------------|
| hf            | -239.97     | kJ/mol | Cheméo Relay (1.0) |
| hfus          | 23.10       | kJ/mol | Joback Method      |
| hvap          | 72.89       | kJ/mol | Cheméo Relay (1.0) |
| ie            | 8.42 ± 0.03 | eV     | NIST Webbook       |
| logp          | 3.137       |        | Crippen Method     |
| mcvol         | 142.260     | ml/mol | McGowan Method     |
| tb            | 527.62      | K      | Cheméo Relay (1.0) |
| tf            | 331.59      | K      | Cheméo Relay (1.0) |

## Temperature Dependent Properties

| Property code | Value     | Unit    | Temperature [K] | Source        |
|---------------|-----------|---------|-----------------|---------------|
| cpg           | 274.61    | J/mol×K | 627.69          | Joback Method |
| cpg           | 318.89    | J/mol×K | 878.22          | Joback Method |
| cpg           | 313.01    | J/mol×K | 836.47          | Joback Method |
| cpg           | 306.56    | J/mol×K | 794.71          | Joback Method |
| cpg           | 299.53    | J/mol×K | 752.96          | Joback Method |
| cpg           | 291.87    | J/mol×K | 711.20          | Joback Method |
| cpg           | 283.58    | J/mol×K | 669.45          | Joback Method |
| dvisc         | 0.0004344 | Pa×s    | 525.42          | Joback Method |
| dvisc         | 0.0002385 | Pa×s    | 627.69          | Joback Method |
| dvisc         | 0.0002847 | Pa×s    | 593.60          | Joback Method |
| dvisc         | 0.0003472 | Pa×s    | 559.51          | Joback Method |
| dvisc         | 0.0010566 | Pa×s    | 423.14          | Joback Method |

|       |           |      |        |               |
|-------|-----------|------|--------|---------------|
| dvisc | 0.0005606 | Pa×s | 491.32 | Joback Method |
| dvisc | 0.0007517 | Pa×s | 457.23 | Joback Method |

## Sources

|                                  |                                                                                                                                               |
|----------------------------------|-----------------------------------------------------------------------------------------------------------------------------------------------|
| <b>Crippen Method:</b>           | <a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>                             |
| <b>Cheméo Relay (1.0):</b>       |                                                                                                                                               |
| <b>Cheméo Relay (1.0, beta):</b> |                                                                                                                                               |
| <b>NIST Webbook:</b>             | <a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C28165722&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C28165722&amp;Units=SI</a> |
| <b>Joback Method:</b>            | <a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>                                         |
| <b>McGowan Method:</b>           | <a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>                         |
| <b>Crippen Method:</b>           | <a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>                                     |

## Legend

|               |                                                 |
|---------------|-------------------------------------------------|
| <b>cpg:</b>   | Ideal gas heat capacity                         |
| <b>dvisc:</b> | Dynamic viscosity                               |
| <b>hf:</b>    | Enthalpy of formation at standard conditions    |
| <b>hfus:</b>  | Enthalpy of fusion at standard conditions       |
| <b>hvap:</b>  | Enthalpy of vaporization at standard conditions |
| <b>ie:</b>    | Ionization energy                               |
| <b>logp:</b>  | Octanol/Water partition coefficient             |
| <b>mcvol:</b> | McGowan's characteristic volume                 |
| <b>tb:</b>    | Normal Boiling Point Temperature                |
| <b>tf:</b>    | Normal melting (fusion) point                   |

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