

# Propane, 2-methyl-1-(propylthio)-

<b>Other names:</b>	Sulfide, isobutyl propyl Isobutyl propyl sulfide 2-Methyl-4-thiaheptane (2-methylpropyl) propyl sulfide Propyl 2-methylpropylsulfide Propyl isobutyl sulfide
<b>Inchi:</b>	InChI=1S/C7H16S/c1-4-5-8-6-7(2)3/h7H,4-6H2,1-3H3
<b>InchiKey:</b>	YZCUJPLKGSDFP-UHFFFAOYSA-N
<b>Formula:</b>	C7H16S
<b>SMILES:</b>	CCCSCC(C)C
<b>Mol. weight [g/mol]:</b>	132.27
<b>CAS:</b>	1741-84-0

## Physical Properties

Property code	Value	Unit	Source
gf	38.74	kJ/mol	Joback Method
hf	-151.22	kJ/mol	Joback Method
hfus	14.49	kJ/mol	Joback Method
hvap	37.61	kJ/mol	Joback Method
ie	8.40 ± 0.05	eV	NIST Webbook
log10ws	-2.39		Crippen Method
logp	2.786		Crippen Method
mcvol	125.840	ml/mol	McGowan Method
pc	2875.03	kPa	Joback Method
rinpole	941.00		NIST Webbook
rinpole	941.00		NIST Webbook
rinpole	941.00		NIST Webbook
rinpole	941.00		NIST Webbook
rinpole	936.00		NIST Webbook
tb	427.90	K	Joback Method
tc	619.57	K	Joback Method
tf	188.05	K	Joback Method
vc	0.475	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	245.63	J/molxK	427.90	Joback Method
cpg	258.76	J/molxK	459.85	Joback Method
cpg	271.36	J/molxK	491.79	Joback Method
cpg	283.45	J/molxK	523.74	Joback Method
cpg	295.02	J/molxK	555.68	Joback Method
cpg	306.08	J/molxK	587.63	Joback Method
cpg	316.66	J/molxK	619.57	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1741840&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1741840&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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