

# Adipic acid, isohexyl 3-oxobut-2-yl ester

<b>Inchi:</b>	InChI=1S/C16H28O5/c1-12(2)8-7-11-20-15(18)9-5-6-10-16(19)21-14(4)13(3)17/h12,14H
<b>InchiKey:</b>	KBIOMBDNDXGHLQ-UHFFFAOYSA-N
<b>Formula:</b>	C16H28O5
<b>SMILES:</b>	CC(=O)C(C)OC(=O)CCCCC(=O)OCCCC(C)C
<b>Mol. weight [g/mol]:</b>	300.39

## Physical Properties

Property code	Value	Unit	Source
gf	-517.80	kJ/mol	Joback Method
hf	-986.31	kJ/mol	Joback Method
hfus	37.32	kJ/mol	Joback Method
hvap	75.49	kJ/mol	Joback Method
log10ws	-3.40		Crippen Method
logp	3.047		Crippen Method
mvol	252.750	ml/mol	McGowan Method
pc	1498.83	kPa	Joback Method
rinpol	2004.00		NIST Webbook
rinpol	2004.00		NIST Webbook
tb	771.05	K	Joback Method
tc	959.08	K	Joback Method
tf	434.33	K	Joback Method
vc	0.974	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	757.14	J/mol×K	771.05	Joback Method
cpg	772.74	J/mol×K	802.39	Joback Method
cpg	787.40	J/mol×K	833.73	Joback Method
cpg	801.14	J/mol×K	865.07	Joback Method
cpg	813.97	J/mol×K	896.40	Joback Method
cpg	825.88	J/mol×K	927.74	Joback Method
cpg	836.89	J/mol×K	959.08	Joback Method
dvisc	0.0013691	Paxs	434.33	Joback Method

dvisc	0.0006419	Paxs	490.45	Joback Method
dvisc	0.0003516	Paxs	546.57	Joback Method
dvisc	0.0002154	Paxs	602.69	Joback Method
dvisc	0.0001435	Paxs	658.81	Joback Method
dvisc	0.0001019	Paxs	714.93	Joback Method
dvisc	0.0000760	Paxs	771.05	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353749&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353749&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/53-488-2/Adipic-acid-isoheptyl-3-oxobut-2-yl-ester.pdf>

Generated by Cheméo on 2024-04-29 00:43:14.548183046 +0000 UTC m=+16640643.468760362.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.