

# 1,3-Isobenzofurandione, 5,5'-carbonylbis-

**Other names:**

3,3',4,4'-Benzophenonetetracarboxylic acid dianhydride  
3,3',4,4'-Tetracarboxybenzophenone dianhydride  
3,3',4,4'-benzophenonetetracarboxylic dianhydride  
4,4'-Carbonylbis(phthalic anhydride)  
4,4'-Diphthalic anhydride ketone  
4,4'-carbonyldiphthalic anhydride  
Benzophenone-3,3',4,4'-tetracarboxylic dianhydride  
Benzophenonetetracarboxylic acid anhydride  
Benzophenonetetracarboxylic acid dianhydride  
Benzophenonetetracarboxylic anhydride  
Benzophenonetetracarboxylic dianhydride  
Bis-(3-phthalyl anhydride) ketone  
NSC 78480  
Phthalic anhydride, 4,4'-carbonyldi-

**Inchi:** InChI=1S/C17H6O7/c18-13(7-1-3-9-11(5-7)16(21)23-14(9)19)8-2-4-10-12(6-8)17(22)24-**InchiKey:** VQVIHDPBMFABCQ-UHFFFAOYSA-N**Formula:** C17H6O7**SMILES:** O=C(c1ccc2c(c1)C(=O)OC2=O)c1ccc2c(c1)C(=O)OC2=O**Mol. weight [g/mol]:** 322.23**CAS:** 2421-28-5

## Physical Properties

Property code	Value	Unit	Source
gf	-376.04	kJ/mol	Joback Method
hf	-708.13	kJ/mol	Joback Method
hfus	36.04	kJ/mol	Joback Method
hvap	93.83	kJ/mol	Joback Method
log10ws	-4.34		Crippen Method
logp	1.539		Crippen Method
mcvol	200.740	ml/mol	McGowan Method
pc	3314.37	kPa	Joback Method
ss	350.20	J/mol×K	NIST Webbook
tb	1063.51	K	Joback Method
tc	1353.54	K	Joback Method
tf	804.58	K	Joback Method
vc	0.763	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	627.03	J/mol×K	1305.20	Joback Method
cpg	627.42	J/mol×K	1063.51	Joback Method
cpg	632.07	J/mol×K	1111.85	Joback Method
cpg	634.41	J/mol×K	1160.19	Joback Method
cpg	634.38	J/mol×K	1208.53	Joback Method
cpg	631.94	J/mol×K	1256.86	Joback Method
cpg	619.58	J/mol×K	1353.54	Joback Method
cps	318.70	J/mol×K	300.00	NIST Webbook

## Sources

Determining the Solubility of Organic Compounds in Supercritical Carbon Dioxide Using a Supercritical Fluid Chromatography Directly Interfaced to a Supercritical Fluid Solubility Apparatus:	<a href="https://www.doi.org/10.1021/acs.jced.6b00081">https://www.doi.org/10.1021/acs.jced.6b00081</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2421285&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2421285&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ss:</b>	Solid phase molar entropy at standard conditions

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/53-514-2/1-3-Isobenzofurandione-5-5-carbonylbis.pdf>

Generated by Cheméo on 2024-04-19 14:36:45.209381927 +0000 UTC m=+15826654.129959243.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.