

# isobutyryl chloride

<b>Other names:</b>	2-Methylpropanoyl chloride
<b>Inchi:</b>	InChI=1S/C4H7ClO/c1-3(2)4(5)6/h3H,1-2H3
<b>InchiKey:</b>	DGMOBVGABMBZSB-UHFFFAOYSA-N
<b>Formula:</b>	C4H7ClO
<b>SMILES:</b>	CC(C)C(=O)Cl
<b>Mol. weight [g/mol]:</b>	106.55
<b>CAS:</b>	79-30-1

## Physical Properties

Property code	Value	Unit	Source
gf	-160.49	kJ/mol	Joback Method
hf	-259.49	kJ/mol	Joback Method
hfus	8.39	kJ/mol	Joback Method
hvap	35.24	kJ/mol	Joback Method
log10ws	-1.18		Crippen Method
logp	1.408		Crippen Method
mcvol	81.030	ml/mol	McGowan Method
pc	4067.32	kPa	Joback Method
rinpol	672.00		NIST Webbook
rinpol	672.00		NIST Webbook
rinpol	672.00		NIST Webbook
tb	365.00 ± 2.00	K	NIST Webbook
tb	333.00 ± 4.00	K	NIST Webbook
tb	363.00 ± 3.00	K	NIST Webbook
tb	365.20	K	NIST Webbook
tc	573.96	K	Joback Method
tf	183.20 ± 0.70	K	NIST Webbook
vc	0.308	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	131.55	J/mol×K	381.78	Joback Method
cpg	138.84	J/mol×K	413.81	Joback Method

cpg	145.81	J/mol×K	445.84	Joback Method
cpg	152.47	J/mol×K	477.87	Joback Method
cpg	158.83	J/mol×K	509.90	Joback Method
cpg	164.89	J/mol×K	541.93	Joback Method
cpg	170.67	J/mol×K	573.96	Joback Method
cpl	131.80	J/mol×K	298.00	NIST Webbook
dvisc	0.0053672	Paxs	199.69	Joback Method
dvisc	0.0025233	Paxs	230.04	Joback Method
dvisc	0.0014145	Paxs	260.39	Joback Method
dvisc	0.0008947	Paxs	290.74	Joback Method
dvisc	0.0006172	Paxs	321.08	Joback Method
dvisc	0.0004539	Paxs	351.43	Joback Method
dvisc	0.0003505	Paxs	381.78	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C79301&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C79301&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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