

# Diethylmalonic acid, isobutyl 2-methoxyethyl ester

Inchi:	InChI=1S/C14H26O5/c1-6-14(7-2,12(15)18-9-8-17-5)13(16)19-10-11(3)4/h11H,6-10H2,1
InchiKey:	SPYJIKSVBPVIQZ-UHFFFAOYSA-N
Formula:	C14H26O5
SMILES:	CCC(CC)(C(=O)OCCOC)C(=O)OCC(C)C
Mol. weight [g/mol]:	274.35

## Physical Properties

Property code	Value	Unit	Source
gf	-505.44	kJ/mol	Joback Method
hf	-968.14	kJ/mol	Joback Method
hfus	27.84	kJ/mol	Joback Method
hvap	65.80	kJ/mol	Joback Method
log10ws	-2.01		Crippen Method
logp	2.182		Crippen Method
mcvol	228.870	ml/mol	McGowan Method
pc	1652.46	kPa	Joback Method
rinpol	1570.00		NIST Webbook
tb	691.05	K	Joback Method
tc	876.32	K	Joback Method
tf	401.51	K	Joback Method
vc	0.869	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	653.91	J/molxK	691.05	Joback Method
cpg	725.75	J/molxK	845.44	Joback Method
cpg	713.06	J/molxK	814.56	Joback Method
cpg	699.53	J/molxK	783.68	Joback Method
cpg	685.17	J/molxK	752.81	Joback Method
cpg	669.96	J/molxK	721.93	Joback Method
cpg	737.63	J/molxK	876.32	Joback Method
dvisc	0.0000720	Paxs	691.05	Joback Method
dvisc	0.0000972	Paxs	642.79	Joback Method

dvisc	0.0001378	Paxs	594.54	Joback Method
dvisc	0.0002078	Paxs	546.28	Joback Method
dvisc	0.0003394	Paxs	498.02	Joback Method
dvisc	0.0006159	Paxs	449.77	Joback Method
dvisc	0.0012897	Paxs	401.51	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370671&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370671&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/53-579-1/Diethylmalonic-acid-isobutyl-2-methoxyethyl-ester.pdf>

Generated by Cheméo on 2024-04-29 19:46:57.854820217 +0000 UTC m=+16709266.775397537.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.