

# Adipic acid, octadecyl 2,2,3,3,4,4,5,5-octafluoropentyl ester

Inchi:	InChI=1S/C29H48F8O4/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-19-22-40-24(38)20-17
InchiKey:	RJXNBVTVZREHGI-UHFFFAOYSA-N
Formula:	C29H48F8O4
SMILES:	CCCCCCCCCCCCCCCCCOC(=O)CCCC(=O)OCC(F)(F)C(F)(F)C(F)(F)C(F)F
Mol. weight [g/mol]:	612.68

## Physical Properties

Property code	Value	Unit	Source
gf	-1826.94	kJ/mol	Joback Method
hf	-2731.90	kJ/mol	Joback Method
hfus	75.31	kJ/mol	Joback Method
hvap	87.65	kJ/mol	Joback Method
log10ws	-10.94		Crippen Method
logp	10.066		Crippen Method
mvol	448.510	ml/mol	McGowan Method
pc	565.55	kPa	Joback Method
rinpol	2979.00		NIST Webbook
rinpol	2979.00		NIST Webbook
tb	999.53	K	Joback Method
tc	1275.61	K	Joback Method
tf	557.89	K	Joback Method
vc	1.812	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1591.75	J/molxK	999.53	Joback Method
cpg	1616.63	J/molxK	1045.54	Joback Method
cpg	1639.47	J/molxK	1091.56	Joback Method
cpg	1660.54	J/molxK	1137.57	Joback Method
cpg	1680.12	J/molxK	1183.58	Joback Method
cpg	1698.48	J/molxK	1229.59	Joback Method
cpg	1715.88	J/molxK	1275.61	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U353743&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U353743&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvp:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinp:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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