

# 3-Methylenecyclohexene

<b>Other names:</b>	Cyclohexene, 3-methylene- 1-Methylene-2-cyclohexene
<b>Inchi:</b>	InChI=1S/C7H10/c1-7-5-3-2-4-6-7/h3,5H,1-2,4,6H2
<b>InchiKey:</b>	XVKFDCVTYBMNRZ-UHFFFAOYSA-N
<b>Formula:</b>	C7H10
<b>SMILES:</b>	C=C1C=CCCC1
<b>Mol. weight [g/mol]:</b>	94.15
<b>CAS:</b>	1888-90-0

## Physical Properties

Property code	Value	Unit	Source
gf	123.26	kJ/mol	Joback Method
hf	28.87	kJ/mol	Joback Method
hfus	4.71	kJ/mol	Joback Method
hvap	32.37	kJ/mol	Joback Method
log10ws	-2.35		Crippen Method
logp	2.283		Crippen Method
mvol	90.030	ml/mol	McGowan Method
pc	3920.94	kPa	Joback Method
tb	382.00 ± 3.00	K	NIST Webbook
tc	589.39	K	Joback Method
tf	194.71	K	Joback Method
vc	0.332	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	149.14	J/mol×K	382.10	Joback Method
cpg	162.07	J/mol×K	416.65	Joback Method
cpg	174.34	J/mol×K	451.20	Joback Method
cpg	185.95	J/mol×K	485.75	Joback Method
cpg	196.93	J/mol×K	520.29	Joback Method
cpg	207.30	J/mol×K	554.84	Joback Method
cpg	217.08	J/mol×K	589.39	Joback Method

dvisc	0.0042247	Paxs	194.71	Joback Method
dvisc	0.0019440	Paxs	225.94	Joback Method
dvisc	0.0010801	Paxs	257.17	Joback Method
dvisc	0.0006816	Paxs	288.41	Joback Method
dvisc	0.0004706	Paxs	319.64	Joback Method
dvisc	0.0003471	Paxs	350.87	Joback Method
dvisc	0.0002690	Paxs	382.10	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1888900&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1888900&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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