

# 2-Bromo-4,6-di-tert-butylphenol

<b>Other names:</b>	Phenol, 2-bromo-4,6-bis(1,1-dimethylethyl)- 2-Bromo-4,6-di-t-butylphenol
<b>Inchi:</b>	InChI=1S/C14H21BrO/c1-13(2,3)9-7-10(14(4,5)6)12(16)11(15)8-9/h7-8,16H,1-6H3
<b>InchiKey:</b>	DIWZVAHZEOFSL5-UHFFFAOYSA-N
<b>Formula:</b>	C14H21BrO
<b>SMILES:</b>	CC(C)(C)c1cc(Br)c(O)c(C(C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	285.22
<b>CAS:</b>	20834-61-1

## Physical Properties

Property code	Value	Unit	Source
gf	25.53	kJ/mol	Joback Method
hf	-287.18	kJ/mol	Joback Method
hfus	21.52	kJ/mol	Joback Method
hvap	67.22	kJ/mol	Joback Method
log10ws	-4.81		Crippen Method
logp	4.750		Crippen Method
mvol	207.730	ml/mol	McGowan Method
pc	2487.55	kPa	Joback Method
tb	696.68	K	Joback Method
tc	938.99	K	Joback Method
tf	475.36	K	Joback Method
vc	0.718	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	551.70	J/molxK	696.68	Joback Method
cpg	567.42	J/molxK	737.06	Joback Method
cpg	582.07	J/molxK	777.45	Joback Method
cpg	595.83	J/molxK	817.83	Joback Method
cpg	608.86	J/molxK	858.22	Joback Method
cpg	621.34	J/molxK	898.60	Joback Method
cpg	633.45	J/molxK	938.99	Joback Method

dvisc	0.0002806	Paxs	475.36	Joback Method
dvisc	0.0001348	Paxs	512.25	Joback Method
dvisc	0.0000715	Paxs	549.13	Joback Method
dvisc	0.0000411	Paxs	586.02	Joback Method
dvisc	0.0000252	Paxs	622.91	Joback Method
dvisc	0.0000163	Paxs	659.79	Joback Method
dvisc	0.0000111	Paxs	696.68	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20834611&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20834611&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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