

# (+)-di-O-4-Toluoyl-D-tartaric acid

**Other names:**

(+)-di-O,O'-p-Toluyl-D-tartaric acid  
di-p-Toluoyl-L-tartaric acid  
(-)-di-p-Toluyl-L-tartaric acid  
(2S,3S)-(-)-di-o-4-Toluoyl-D-tartaric acid  
L-di-p-Toluyltartaric acid  
Butanedioic acid, 2,3-bis[(4-methylbenzoyl)oxy]-, [S-(R\*,R\*)]-  
[S(R\*,R\*)]-2,3-bis[(4-methylbenzoyl)oxy]succinic acid

**Inchi:**

InChI=1S/C20H18O8/c1-11-3-7-13(8-4-11)19(25)27-15(17(21)22)16(18(23)24)28-20(26)

**InchiKey:**

CMIBUZBMZCBCAT-UHFFFAOYSA-N

**Formula:**

C<sub>20</sub>H<sub>18</sub>O<sub>8</sub>

**SMILES:**

Cc1ccc(C(=O)OC(C(=O)O)C(OC(=O)c2ccc(C)cc2)C(=O)O)cc1

**Mol. weight [g/mol]:**

386.35

**CAS:**

32634-68-7

## Physical Properties

Property code	Value	Unit	Source
gf	-681.12	kJ/mol	Joback Method
hf	-1035.79	kJ/mol	Joback Method
hfus	44.76	kJ/mol	Joback Method
hvap	130.38	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	2.224		Crippen Method
mvol	274.900	ml/mol	McGowan Method
pc	2293.71	kPa	Joback Method
tb	1164.12	K	Joback Method
tc	1427.36	K	Joback Method
tf	728.86	K	Joback Method
vc	1.026	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	867.78	J/mol×K	1164.12	Joback Method
cpg	871.48	J/mol×K	1207.99	Joback Method

cpg	873.65	J/molxK	1251.87	Joback Method
cpg	874.34	J/molxK	1295.74	Joback Method
cpg	873.60	J/molxK	1339.61	Joback Method
cpg	871.49	J/molxK	1383.49	Joback Method
cpg	868.05	J/molxK	1427.36	Joback Method
dvisc	0.0000152	Paxs	728.86	Joback Method
dvisc	0.0000065	Paxs	801.40	Joback Method
dvisc	0.0000032	Paxs	873.95	Joback Method
dvisc	0.0000017	Paxs	946.49	Joback Method
dvisc	0.0000010	Paxs	1019.03	Joback Method
dvisc	0.0000007	Paxs	1091.58	Joback Method
dvisc	0.0000004	Paxs	1164.12	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C32634687&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C32634687&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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