

# Pentanoic acid, 2,4-dioxo-, methyl ester

<b>Other names:</b>	methyl acetopyruvate
<b>Inchi:</b>	InChI=1S/C6H8O4/c1-4(7)3-5(8)6(9)10-2/h3H2,1-2H3
<b>InchiKey:</b>	OMHOEQINEXASKE-UHFFFAOYSA-N
<b>Formula:</b>	C6H8O4
<b>SMILES:</b>	<chem>COC(=O)C(=O)CC(C)=O</chem>
<b>Mol. weight [g/mol]:</b>	144.13
<b>CAS:</b>	20577-61-1

## Physical Properties

Property code	Value	Unit	Source
gf	-492.12	kJ/mol	Joback Method
hf	-637.13	kJ/mol	Joback Method
hfus	17.28	kJ/mol	Joback Method
hvap	51.60	kJ/mol	Joback Method
log10ws	0.24		Crippen Method
logp	-0.292		Crippen Method
mcvol	105.980	ml/mol	McGowan Method
pc	3801.00	kPa	Joback Method
tb	520.71	K	Joback Method
tc	719.19	K	Joback Method
tf	329.40	K	Joback Method
vc	0.407	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	227.36	J/mol×K	520.71	Joback Method
cpg	236.04	J/mol×K	553.79	Joback Method
cpg	244.33	J/mol×K	586.87	Joback Method
cpg	252.25	J/mol×K	619.95	Joback Method
cpg	259.79	J/mol×K	653.03	Joback Method
cpg	266.93	J/mol×K	686.11	Joback Method
cpg	273.68	J/mol×K	719.19	Joback Method
dvisc	0.0023960	Paxs	329.40	Joback Method

dvisc	0.0015048	Paxs	361.29	Joback Method
dvisc	0.0010192	Paxs	393.17	Joback Method
dvisc	0.0007318	Paxs	425.06	Joback Method
dvisc	0.0005503	Paxs	456.94	Joback Method
dvisc	0.0004295	Paxs	488.83	Joback Method
dvisc	0.0003456	Paxs	520.71	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20577611&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20577611&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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