

# Diazene, bis-(2-methylpropyl)

<b>Other names:</b>	Diisobutyl diazene
<b>Inchi:</b>	InChI=1S/C8H18N2/c1-7(2)5-9-10-6-8(3)4/h7-8H,5-6H2,1-4H3
<b>InchiKey:</b>	JUOJXNAVZADLAJ-UHFFFAOYSA-N
<b>Formula:</b>	C8H18N2
<b>SMILES:</b>	CC(C)CN=NCC(C)C
<b>Mol. weight [g/mol]:</b>	142.24
<b>CAS:</b>	3896-19-3

## Physical Properties

Property code	Value	Unit	Source
hf	-171.79	kJ/mol	Joback Method
hvap	39.30	kJ/mol	Joback Method
ie	8.20	eV	NIST Webbook
log10ws	-1.95		Crippen Method
logp	2.751		Crippen Method
mcvol	139.240	ml/mol	McGowan Method
pc	2007.30	kPa	Joback Method
rinpol	867.00		NIST Webbook
tb	530.76	K	Joback Method
tc	733.01	K	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	324.00 ± 1.00	K	3.20	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3896193&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3896193&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature

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