

# 4-(para-Tolyl)-butyric acid

<b>Other names:</b>	4-(p-Tolyl)butyric acid Benzenebutanoic acid, 4-methyl-
<b>Inchi:</b>	InChI=1S/C11H14O2/c1-9-5-7-10(8-6-9)3-2-4-11(12)13/h5-8H,2-4H2,1H3,(H,12,13)
<b>InchiKey:</b>	IXWOVMRDYFFXGI-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O2
<b>SMILES:</b>	Cc1ccc(CCCC(=O)O)cc1
<b>Mol. weight [g/mol]:</b>	178.23
<b>CAS:</b>	4521-22-6

## Physical Properties

Property code	Value	Unit	Source
gf	-121.22	kJ/mol	Joback Method
hf	-310.12	kJ/mol	Joback Method
hfus	23.59	kJ/mol	Joback Method
hvap	66.44	kJ/mol	Joback Method
log10ws	-2.69		Crippen Method
logp	2.402		Crippen Method
mvol	149.530	ml/mol	McGowan Method
pc	3076.16	kPa	Joback Method
tb	628.79	K	Joback Method
tc	827.28	K	Joback Method
tf	328.15 ± 2.00	K	NIST Webbook
vc	0.569	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	374.63	J/mol×K	628.79	Joback Method
cpg	427.19	J/mol×K	794.20	Joback Method
cpg	417.93	J/mol×K	761.12	Joback Method
cpg	408.07	J/mol×K	728.04	Joback Method
cpg	397.58	J/mol×K	694.95	Joback Method
cpg	386.44	J/mol×K	661.87	Joback Method
cpg	435.88	J/mol×K	827.28	Joback Method

dvisc	0.0000758	Paxs	628.79	Joback Method
dvisc	0.0001127	Paxs	584.56	Joback Method
dvisc	0.0001788	Paxs	540.33	Joback Method
dvisc	0.0003078	Paxs	496.11	Joback Method
dvisc	0.0005896	Paxs	451.88	Joback Method
dvisc	0.0013004	Paxs	407.65	Joback Method
dvisc	0.0034768	Paxs	363.42	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C4521226&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C4521226&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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