

# 4-Butylbenzoic acid, 2-adamantyl ester

<b>Inchi:</b>	InChI=1S/C21H28O2/c1-2-3-4-14-5-7-17(8-6-14)21(22)23-20-18-10-15-9-16(12-18)13-19
<b>InchiKey:</b>	XPXKHQKPWRXECF-UHFFFAOYSA-N
<b>Formula:</b>	C21H28O2
<b>SMILES:</b>	CCCCc1ccc(C(=O)OC2C3CC4CC(C3)CC2C4)cc1
<b>Mol. weight [g/mol]:</b>	312.45

## Physical Properties

Property code	Value	Unit	Source
gf	149.53	kJ/mol	Joback Method
hf	-324.95	kJ/mol	Joback Method
hfus	41.03	kJ/mol	Joback Method
hvap	73.73	kJ/mol	Joback Method
log10ws	-5.95		Crippen Method
logp	5.011		Crippen Method
mcvol	257.850	ml/mol	McGowan Method
pc	1570.96	kPa	Joback Method
rinpol	2462.40		NIST Webbook
tb	802.98	K	Joback Method
tc	1025.82	K	Joback Method
tf	479.35	K	Joback Method
vc	0.989	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	848.28	J/molxK	802.98	Joback Method
cpg	868.71	J/molxK	840.12	Joback Method
cpg	887.81	J/molxK	877.26	Joback Method
cpg	905.71	J/molxK	914.40	Joback Method
cpg	922.52	J/molxK	951.54	Joback Method
cpg	938.35	J/molxK	988.68	Joback Method
cpg	953.34	J/molxK	1025.82	Joback Method
dvisc	0.0039208	Paxs	479.35	Joback Method
dvisc	0.0033058	Paxs	533.29	Joback Method

dvisc	0.0028759	Paxs	587.23	Joback Method
dvisc	0.0025613	Paxs	641.16	Joback Method
dvisc	0.0023225	Paxs	695.10	Joback Method
dvisc	0.0021359	Paxs	749.04	Joback Method
dvisc	0.0019864	Paxs	802.98	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292207&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292207&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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