

# 2,4,8,10-Tetraoxaspiro[5.5]undecane

<b>Other names:</b>	Formaldehyde, cyclic diacetal with pentaerythritol Pentaerythritol bisformal Pentaerythritol cyclic diformal Pentaerythritol diformal Pentaerythritol, bis(cyclic acetal) with formaldehyde- Pentaerythritol, dimethylene- 5,5'-Spirobi-m-dioxane 5,5'-Spirobi-1,3-dioxane NSC 139455
<b>Inchi:</b>	InChI=1S/C7H12O4/c1-7(2-9-5-8-1)3-10-6-11-4-7/h1-6H2
<b>InchiKey:</b>	BGCSUUSPRCDKBQ-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O4
<b>SMILES:</b>	C1OCC2(CO1)COCOC2
<b>Mol. weight [g/mol]:</b>	160.17
<b>CAS:</b>	126-54-5

## Physical Properties

Property code	Value	Unit	Source
chs	-3767.00	kJ/mol	NIST Webbook
gf	-273.20	kJ/mol	Joback Method
hf	-565.43	kJ/mol	Joback Method
hfus	24.20	kJ/mol	Joback Method
hsub	73.00 ± 2.00	kJ/mol	NIST Webbook
hvap	49.06	kJ/mol	Joback Method
log10ws	0.36		Crippen Method
logp	-0.019		Crippen Method
mcvol	111.250	ml/mol	McGowan Method
pc	4672.09	kPa	Joback Method
tb	507.10	K	Joback Method
tc	753.39	K	Joback Method
tf	321.35	K	Joback Method
vc	0.385	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.52	J/mol×K	507.10	Joback Method
cpg	295.57	J/mol×K	548.15	Joback Method
cpg	311.20	J/mol×K	589.20	Joback Method
cpg	325.60	J/mol×K	630.25	Joback Method
cpg	338.92	J/mol×K	671.29	Joback Method
cpg	351.34	J/mol×K	712.34	Joback Method
cpg	363.04	J/mol×K	753.39	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	354.70	K	0.10	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C126545&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C126545&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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