

# 8-methyl-tetracosane

<b>Other names:</b>	Tetracosane, 8-methyl
<b>Inchi:</b>	InChI=1S/C25H52/c1-4-6-8-10-11-12-13-14-15-16-17-18-20-22-24-25(3)23-21-19-9-7-5-2
<b>InchiKey:</b>	MXKSWGQIAVWJOI-UHFFFAOYSA-N
<b>Formula:</b>	C25H52
<b>SMILES:</b>	CCCCCCCCCCCCCCCC(C)CCCCCCC
<b>Mol. weight [g/mol]:</b>	352.68

## Physical Properties

Property code	Value	Unit	Source
gf	157.18	kJ/mol	Joback Method
hf	-564.61	kJ/mol	Joback Method
hfus	56.98	kJ/mol	Joback Method
hvap	70.86	kJ/mol	Joback Method
log10ws	-10.05		Crippen Method
logp	9.854		Crippen Method
mcvol	363.110	ml/mol	McGowan Method
pc	767.34	kPa	Joback Method
rinpol	2441.50		NIST Webbook
rinpol	2434.00		NIST Webbook
rinpol	2434.00		NIST Webbook
tb	770.96	K	Joback Method
tc	944.71	K	Joback Method
tf	356.51	K	Joback Method
vc	1.429	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1151.89	J/molxK	770.96	Joback Method
cpg	1175.34	J/molxK	799.92	Joback Method
cpg	1197.69	J/molxK	828.88	Joback Method
cpg	1218.99	J/molxK	857.83	Joback Method
cpg	1239.27	J/molxK	886.79	Joback Method
cpg	1258.59	J/molxK	915.75	Joback Method

cpg	1276.97	J/molxK	944.71	Joback Method
dvisc	0.0027240	Paxs	356.51	Joback Method
dvisc	0.0007960	Paxs	425.58	Joback Method
dvisc	0.0003280	Paxs	494.66	Joback Method
dvisc	0.0001679	Paxs	563.73	Joback Method
dvisc	0.0000995	Paxs	632.81	Joback Method
dvisc	0.0000654	Paxs	701.88	Joback Method
dvisc	0.0000463	Paxs	770.96	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R404745&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R404745&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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