

# Butane, 1,1'-sulfinylbis-

<b>Other names:</b>	1,1'-sulfinylbisbutane Di-n-butyl sulfoxide Dibutyl sulphoxide Sulfoxide, dibutyl butyl sulfoxide dibutyl sulfoxide n-Butyl sulfoxide
<b>Inchi:</b>	InChI=1S/C8H18OS/c1-3-5-7-10(9)8-6-4-2/h3-8H2,1-2H3
<b>InchiKey:</b>	LOWMYOWHQMKBTM-UHFFFAOYSA-N
<b>Formula:</b>	C8H18OS
<b>SMILES:</b>	CCCCS(=O)CCCC
<b>Mol. weight [g/mol]:</b>	162.29
<b>CAS:</b>	2168-93-6

## Physical Properties

Property code	Value	Unit	Source
gf	-201.23	kJ/mol	Joback Method
hf	-414.19	kJ/mol	Joback Method
hfus	24.23	kJ/mol	Joback Method
hvap	46.13	kJ/mol	Joback Method
log10ws	-1.81		Crippen Method
logp	2.335		Crippen Method
mcvol	145.800	ml/mol	McGowan Method
pc	2770.08	kPa	Joback Method
tb	440.72	K	Joback Method
tc	614.01	K	Joback Method
tf	305.00 ± 4.00	K	NIST Webbook
tf	306.00 ± 5.00	K	NIST Webbook
tf	305.00 ± 4.00	K	NIST Webbook
vc	0.574	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	298.51	J/mol×K	440.72	Joback Method
cpg	312.25	J/mol×K	469.60	Joback Method
cpg	325.47	J/mol×K	498.48	Joback Method
cpg	338.19	J/mol×K	527.37	Joback Method
cpg	350.42	J/mol×K	556.25	Joback Method
cpg	362.16	J/mol×K	585.13	Joback Method
cpg	373.41	J/mol×K	614.01	Joback Method
srf	0.03	N/m	313.15	Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures

## Sources

Surface and bulk behavior of (dialkylsulfoxides + carbon tetrachloride) mixtures:

McGowan Method:

NIST Webbook:

Crippen Method:

Crippen Method:

<https://www.doi.org/10.1016/j.jct.2009.06.021>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2168936&Units=SI>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>srf:</b>	Surface Tension
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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