

# Phthalic acid, monoethyl ester

<b>Other names:</b>	1,2-Benzenecarboxylic acid, monoethyl ester Monoethyl phthalate 2-(Ethoxycarbonyl)benzoic acid
<b>Inchi:</b>	InChI=1S/C10H10O4/c1-2-14-10(13)8-6-4-3-5-7(8)9(11)12/h3-6H,2H2,1H3,(H,11,12)
<b>InchiKey:</b>	YWWHKOHZGJFMIE-UHFFFAOYSA-N
<b>Formula:</b>	C10H10O4
<b>SMILES:</b>	CCOC(=O)c1ccccc1C(=O)O
<b>Mol. weight [g/mol]:</b>	194.18
<b>CAS:</b>	2306-33-4

## Physical Properties

Property code	Value	Unit	Source
gf	-363.56	kJ/mol	Joback Method
hf	-534.28	kJ/mol	Joback Method
hfus	23.78	kJ/mol	Joback Method
hvap	73.37	kJ/mol	Joback Method
log10ws	-2.19		Crippen Method
logp	1.561		Crippen Method
mcvol	142.880	ml/mol	McGowan Method
pc	3624.61	kPa	Joback Method
rinpol	1651.00		NIST Webbook
rinpol	1651.00		NIST Webbook
tb	682.20	K	Joback Method
tc	888.42	K	Joback Method
tf	424.31	K	Joback Method
vc	0.536	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	361.74	J/molxK	682.20	Joback Method
cpg	404.35	J/molxK	854.05	Joback Method
cpg	397.06	J/molxK	819.68	Joback Method
cpg	389.16	J/molxK	785.31	Joback Method

cpg	380.65	J/molxK	750.94	Joback Method
cpg	371.51	J/molxK	716.57	Joback Method
cpg	411.06	J/molxK	888.42	Joback Method
dvisc	0.0000584	Paxs	682.20	Joback Method
dvisc	0.0000831	Paxs	639.22	Joback Method
dvisc	0.0001242	Paxs	596.24	Joback Method
dvisc	0.0001977	Paxs	553.25	Joback Method
dvisc	0.0003404	Paxs	510.27	Joback Method
dvisc	0.0006477	Paxs	467.29	Joback Method
dvisc	0.0014037	Paxs	424.31	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2306334&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2306334&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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