

Heneicosanoic acid

Other names:	henicosanoic acid n-Heneicosanoic acid
Inchi:	InChI=1S/C21H42O2/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21(22)23/h2
InchiKey:	CKDDRHZIAZRDBW-UHFFFAOYSA-N
Formula:	C21H42O2
SMILES:	CCCCCCCCCCCCCCCCCCCC(=O)O
Mol. weight [g/mol]:	326.56
CAS:	2363-71-5

Physical Properties

Property code	Value	Unit	Source
gf	-139.80	kJ/mol	Joback Method
hf	-741.58	kJ/mol	Joback Method
hfus	55.83	kJ/mol	Joback Method
hvap	85.77	kJ/mol	Joback Method
log10ws	-7.71		Crippen Method
logp	7.503		Crippen Method
mvol	314.190	ml/mol	McGowan Method
pc	1054.83	kPa	Joback Method
rinpol	2463.20		NIST Webbook
tb	825.93	K	Joback Method
tc	1011.18	K	Joback Method
tf	347.65 ± 2.00	K	NIST Webbook
vc	1.236	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1101.16	J/molxK	980.30	Joback Method
cpg	1115.55	J/molxK	1011.18	Joback Method
cpg	1016.16	J/molxK	825.93	Joback Method
cpg	1035.04	J/molxK	856.80	Joback Method
cpg	1052.93	J/molxK	887.68	Joback Method
cpg	1069.89	J/molxK	918.55	Joback Method

cpg	1085.95	J/molxK	949.43	Joback Method
dvisc	0.0000141	Paxs	825.93	Joback Method
dvisc	0.0000218	Paxs	761.14	Joback Method
dvisc	0.0013730	Paxs	437.18	Joback Method
dvisc	0.0003911	Paxs	501.97	Joback Method
dvisc	0.0001484	Paxs	566.76	Joback Method
dvisc	0.0000687	Paxs	631.55	Joback Method
dvisc	0.0000367	Paxs	696.35	Joback Method
hfust	63.00	kJ/mol	346.70	NIST Webbook
hvapt	149.20	kJ/mol	298.00	Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of Some Fatty Acids

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.87163e+01
Coeff. B	-7.56794e+03
Coeff. C	-1.38099e+02
Temperature range (K), min.	548.76
Temperature range (K), max.	702.67

Sources

Vapor Pressures and Vaporization, Sublimation, and Fusion Enthalpies of Some Fatty Acids:

McGowan Method:

NIST Webbook:

The Yaws Handbook of Vapor Pressure:

Crippen Method:

Crippen Method:

<https://www.doi.org/10.1021/je300902c>

https://en.wikipedia.org/wiki/Joback_method

<http://link.springer.com/article/10.1007/BF02311772>

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2363715&Units=SI>

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hfust:	Enthalpy of fusion at a given temperature
hvap:	Enthalpy of vaporization at standard conditions
hvapt:	Enthalpy of vaporization at a given temperature
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
pvap:	Vapor pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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