

# Homojasmonic acid, Ile conjugate, methyl ester

<b>Inchi:</b>	InChI=1S/C20H33NO4/c1-5-7-8-9-16-15(10-12-17(16)22)11-13-18(23)21-19(14(3)6-2)20
<b>InchiKey:</b>	KJUCNIXREXYKMO-QURVCBRASA-N
<b>Formula:</b>	C20H33NO4
<b>SMILES:</b>	CCC=CCC1C(=O)CCC1CCC(=O)NC(C(=O)OC)C(C)CC
<b>Mol. weight [g/mol]:</b>	351.48

## Physical Properties

Property code	Value	Unit	Source
gf	-174.34	kJ/mol	Joback Method
hf	-750.94	kJ/mol	Joback Method
hfus	44.71	kJ/mol	Joback Method
hvap	85.83	kJ/mol	Joback Method
log10ws	-4.44		Crippen Method
logp	3.422		Crippen Method
mvol	298.060	ml/mol	McGowan Method
pc	1307.06	kPa	Joback Method
rinpol	2434.00		NIST Webbook
rinpol	2434.00		NIST Webbook
tb	919.04	K	Joback Method
tc	1133.58	K	Joback Method
tf	529.71	K	Joback Method
vc	1.135	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1008.74	J/mol×K	919.04	Joback Method
cpg	1025.30	J/mol×K	954.80	Joback Method
cpg	1040.41	J/mol×K	990.55	Joback Method
cpg	1054.08	J/mol×K	1026.31	Joback Method
cpg	1066.36	J/mol×K	1062.07	Joback Method
cpg	1077.27	J/mol×K	1097.83	Joback Method
cpg	1086.84	J/mol×K	1133.58	Joback Method

# Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R539198&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R539198&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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