

# 1,4-Benzenedicarboxaldehyde, 2-methyl-

<b>Other names:</b>	Terephthalaldehyde, methyl- 2-Methylterephthalaldehyde 2-Methyl-p-phthalaldehyde
<b>Inchi:</b>	InChI=1S/C9H8O2/c1-7-4-8(5-10)2-3-9(7)6-11/h2-6H,1H3
<b>InchiKey:</b>	MNHWRUCVFATHDL-UHFFFAOYSA-N
<b>Formula:</b>	C9H8O2
<b>SMILES:</b>	<chem>Cc1cc(C=O)ccc1C=O</chem>
<b>Mol. weight [g/mol]:</b>	148.16
<b>CAS:</b>	27587-17-3

## Physical Properties

Property code	Value	Unit	Source
gf	-80.99	kJ/mol	Joback Method
hf	-186.66	kJ/mol	Joback Method
hfus	16.91	kJ/mol	Joback Method
hvap	52.67	kJ/mol	Joback Method
log10ws	-2.43		Crippen Method
logp	1.620		Crippen Method
mcvol	117.050	ml/mol	McGowan Method
pc	3749.97	kPa	Joback Method
tb	539.28	K	Joback Method
tc	758.36	K	Joback Method
tf	326.65	K	Joback Method
vc	0.466	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	250.21	J/molxK	539.28	Joback Method
cpg	260.57	J/molxK	575.79	Joback Method
cpg	270.31	J/molxK	612.31	Joback Method
cpg	279.45	J/molxK	648.82	Joback Method
cpg	288.01	J/molxK	685.33	Joback Method
cpg	296.01	J/molxK	721.84	Joback Method

cpg	303.47	J/mol×K	758.36	Joback Method
dvisc	0.0020171	Paxs	326.65	Joback Method
dvisc	0.0012969	Paxs	362.09	Joback Method
dvisc	0.0009022	Paxs	397.53	Joback Method
dvisc	0.0006660	Paxs	432.96	Joback Method
dvisc	0.0005148	Paxs	468.40	Joback Method
dvisc	0.0004126	Paxs	503.84	Joback Method
dvisc	0.0003404	Paxs	539.28	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27587173&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27587173&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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