

# Propanoic acid, 2-chloro, 1-methyl-3-butenyl ester

Inchi:	InChI=1S/C8H13ClO2/c1-4-5-6(2)11-8(10)7(3)9/h4,6-7H,1,5H2,2-3H3
InchiKey:	QVYJRALGDILKPA-UHFFFAOYSA-N
Formula:	C8H13ClO2
SMILES:	C=CCC(C)OC(=O)C(C)Cl
Mol. weight [g/mol]:	176.64

## Physical Properties

Property code	Value	Unit	Source
gf	-146.41	kJ/mol	Joback Method
hf	-354.12	kJ/mol	Joback Method
hfus	15.13	kJ/mol	Joback Method
hvap	45.50	kJ/mol	Joback Method
log10ws	-2.26		Crippen Method
logp	2.121		Crippen Method
mcvol	138.960	ml/mol	McGowan Method
pc	2715.50	kPa	Joback Method
rinpol	1057.00		NIST Webbook
rinpol	1040.00		NIST Webbook
rinpol	1054.00		NIST Webbook
rinpol	1032.00		NIST Webbook
rinpol	1062.00		NIST Webbook
ripol	1436.00		NIST Webbook
ripol	1434.00		NIST Webbook
ripol	1438.00		NIST Webbook
ripol	1436.00		NIST Webbook
tb	491.96	K	Joback Method
tc	684.44	K	Joback Method
tf	250.24	K	Joback Method
vc	0.525	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	290.45	J/mol×K	491.96	Joback Method

cpg	344.52	J/mol×K	652.36	Joback Method
cpg	334.74	J/mol×K	620.28	Joback Method
cpg	324.45	J/mol×K	588.20	Joback Method
cpg	313.65	J/mol×K	556.12	Joback Method
cpg	302.32	J/mol×K	524.04	Joback Method
cpg	353.80	J/mol×K	684.44	Joback Method
dvisc	0.0002384	Paxs	491.96	Joback Method
dvisc	0.0003194	Paxs	451.67	Joback Method
dvisc	0.0004533	Paxs	411.39	Joback Method
dvisc	0.0006939	Paxs	371.10	Joback Method
dvisc	0.0011784	Paxs	330.81	Joback Method
dvisc	0.0023178	Paxs	290.53	Joback Method
dvisc	0.0056683	Paxs	250.24	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R113756&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R113756&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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