

# Coniferyl aldehyde

<b>Other names:</b>	2-Propenal, 3-(4-hydroxy-3-methoxyphenyl)- 3-Methoxy-4-hydroxycinnamaldehyde 4-Hydroxy-3-methoxycinnamaldehyde Coniferaldehyde Coniferylic aldehyde Cinnamaldehyde, 4-hydroxy-3-methoxy- CONIFERYL ALDEHYDE-1 2-Propenal, 3-(4-hydroxy-3-methoxyphenyl), (E)- 3-methoxy-4-hydroxycinnamaldehyde (coniferilic aldehyde) 4'-hydroxy-3'-methoxycinnamaldehyde
<b>Inchi:</b>	InChI=1S/C10H10O3/c1-13-10-7-8(3-2-6-11)4-5-9(10)12/h2-7,12H,1H3/b3-2+
<b>InchiKey:</b>	DKZBBWMURDFHNE-NSCUHMNNSA-N
<b>Formula:</b>	C10H10O3
<b>SMILES:</b>	<chem>COc1cc(C=CC=O)ccc1O</chem>
<b>Mol. weight [g/mol]:</b>	178.18
<b>CAS:</b>	458-36-6

## Physical Properties

Property code	Value	Unit	Source
gf	-142.82	kJ/mol	Joback Method
hf	-302.56	kJ/mol	Joback Method
hfus	24.77	kJ/mol	Joback Method
hvap	62.89	kJ/mol	Joback Method
log10ws	-1.66		Crippen Method
logp	1.613		Crippen Method
mcvol	137.010	ml/mol	McGowan Method
pc	3886.79	kPa	Joback Method
rinpol	1676.00		NIST Webbook
rinpol	1751.80		NIST Webbook
rinpol	1672.00		NIST Webbook
rinpol	1672.00		NIST Webbook
rinpol	1741.00		NIST Webbook
rinpol	1741.00		NIST Webbook
ripol	3038.00		NIST Webbook
tb	615.72	K	Joback Method
tc	844.66	K	Joback Method
tf	412.27	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	329.26	J/molxK	615.72	Joback Method
cpg	377.06	J/molxK	806.51	Joback Method
cpg	368.65	J/molxK	768.35	Joback Method
cpg	359.75	J/molxK	730.19	Joback Method
cpg	350.26	J/molxK	692.03	Joback Method
cpg	340.13	J/molxK	653.88	Joback Method
cpg	385.04	J/molxK	844.66	Joback Method
dvisc	0.0000290	Paxs	615.72	Joback Method
dvisc	0.0000425	Paxs	581.81	Joback Method
dvisc	0.0000654	Paxs	547.90	Joback Method
dvisc	0.0001063	Paxs	514.00	Joback Method
dvisc	0.0001852	Paxs	480.09	Joback Method
dvisc	0.0003509	Paxs	446.18	Joback Method
dvisc	0.0007386	Paxs	412.27	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C458366&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C458366&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>r<sub>ipol</sub>:</b>	Polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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