

# 1,1'-Biphenyl,3,3'-difluoro-

<b>Other names:</b>	3,3'-difluoro-1,1'-biphenyl
<b>Inchi:</b>	InChI=1S/C12H8F2/c13-11-5-1-3-9(7-11)10-4-2-6-12(14)8-10/h1-8H
<b>InchiKey:</b>	GAYJHUJLHJWCTH-UHFFFAOYSA-N
<b>Formula:</b>	C12H8F2
<b>SMILES:</b>	Fc1cccc(-c2cccc(F)c2)c1
<b>Mol. weight [g/mol]:</b>	190.19
<b>CAS:</b>	396-64-5

## Physical Properties

Property code	Value	Unit	Source
gf	-133.90	kJ/mol	Joback Method
hf	-233.11	kJ/mol	Joback Method
hfus	20.30	kJ/mol	Joback Method
hvap	46.55	kJ/mol	Joback Method
ie	8.35 ± 0.02	eV	NIST Webbook
log10ws	-4.74		Crippen Method
logp	3.632		Crippen Method
mcvol	135.960	ml/mol	McGowan Method
pc	2992.59	kPa	Joback Method
tb	535.82	K	Joback Method
tc	762.30	K	Joback Method
tf	304.06	K	Joback Method
vc	0.527	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	294.88	J/mol×K	535.82	Joback Method
cpg	308.80	J/mol×K	573.57	Joback Method
cpg	321.75	J/mol×K	611.31	Joback Method
cpg	333.77	J/mol×K	649.06	Joback Method
cpg	344.91	J/mol×K	686.81	Joback Method
cpg	355.21	J/mol×K	724.55	Joback Method
cpg	364.72	J/mol×K	762.30	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	403.20	K	1.90	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C396645&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C396645&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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