

# Benzeneacetic acid, 3-methoxy-

<b>Other names:</b>	Acetic acid, (m-methoxyphenyl)- (m-Methoxyphenyl)acetic acid (3-Methoxyphenyl)acetic acid 3-Methoxybenzeneacetic acid
<b>Inchi:</b>	InChI=1S/C9H10O3/c1-12-8-4-2-3-7(5-8)6-9(10)11/h2-5H,6H2,1H3,(H,10,11)
<b>InchiKey:</b>	LEGPZHPSIPPYIO-UHFFFAOYSA-N
<b>Formula:</b>	C9H10O3
<b>SMILES:</b>	COc1cccc(CC(=O)O)c1
<b>Mol. weight [g/mol]:</b>	166.17
<b>CAS:</b>	1798-09-0

## Physical Properties

Property code	Value	Unit	Source
gf	-243.06	kJ/mol	Joback Method
hf	-401.06	kJ/mol	Joback Method
hfus	19.59	kJ/mol	Joback Method
hvap	64.40	kJ/mol	Joback Method
log10ws	-1.49		Crippen Method
logp	1.322		Crippen Method
mcvol	127.220	ml/mol	McGowan Method
pc	3773.04	kPa	Joback Method
rinpol	1473.00		NIST Webbook
tb	605.45	K	Joback Method
tc	807.67	K	Joback Method
tf	363.11	K	Joback Method
vc	0.474	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.01	J/molxK	605.45	Joback Method
cpg	348.17	J/molxK	773.97	Joback Method
cpg	340.24	J/molxK	740.27	Joback Method
cpg	331.76	J/molxK	706.56	Joback Method

cpg	322.74	J/mol×K	672.86	Joback Method
cpg	313.16	J/mol×K	639.15	Joback Method
cpg	355.59	J/mol×K	807.67	Joback Method
dvisc	0.0000785	Paxs	605.45	Joback Method
dvisc	0.0001149	Paxs	565.06	Joback Method
dvisc	0.0001784	Paxs	524.67	Joback Method
dvisc	0.0002980	Paxs	484.28	Joback Method
dvisc	0.0005464	Paxs	443.89	Joback Method
dvisc	0.0011315	Paxs	403.50	Joback Method
dvisc	0.0027550	Paxs	363.11	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1798090&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1798090&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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