

# Benzene, 1,1'-(1,2-cyclobutanediyl)bis-, trans-

<b>Other names:</b>	trans-1,2-Diphenylcyclobutane (2-Phenylcyclobutyl)benzene,(E)-
<b>Inchi:</b>	InChI=1S/C16H16/c1-3-7-13(8-4-1)15-11-12-16(15)14-9-5-2-6-10-14/h1-10,15-16H,11-1
<b>InchiKey:</b>	AERGGMDNGDDGPI-HZPDHXFCSA-N
<b>Formula:</b>	C16H16
<b>SMILES:</b>	<chem>c1ccc(C2CCC2c2ccccc2)cc1</chem>
<b>Mol. weight [g/mol]:</b>	208.30
<b>CAS:</b>	20071-09-4

## Physical Properties

Property code	Value	Unit	Source
gf	349.60	kJ/mol	Joback Method
hf	145.79	kJ/mol	Joback Method
hfus	22.38	kJ/mol	Joback Method
hvap	55.54	kJ/mol	Joback Method
ie	8.20 ± 0.10	eV	NIST Webbook
log10ws	-4.55		Crippen Method
logp	4.348		Crippen Method
mcvol	177.920	ml/mol	McGowan Method
pc	2576.72	kPa	Joback Method
rinpol	1720.00		NIST Webbook
tb	625.18	K	Joback Method
tc	881.09	K	Joback Method
tf	333.10	K	Joback Method
vc	0.663	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	461.63	J/molxK	625.18	Joback Method
cpg	550.39	J/molxK	838.44	Joback Method
cpg	535.66	J/molxK	795.79	Joback Method
cpg	519.54	J/molxK	753.14	Joback Method
cpg	501.91	J/molxK	710.48	Joback Method

cpg	482.65	J/molxK	667.83	Joback Method
cpg	563.88	J/molxK	881.09	Joback Method
dvisc	0.0003289	Paxs	625.18	Joback Method
dvisc	0.0003938	Paxs	576.50	Joback Method
dvisc	0.0004875	Paxs	527.82	Joback Method
dvisc	0.0006302	Paxs	479.14	Joback Method
dvisc	0.0008633	Paxs	430.46	Joback Method
dvisc	0.0012816	Paxs	381.78	Joback Method
dvisc	0.0021353	Paxs	333.10	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C20071094&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C20071094&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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