

# Propanedioic acid, (phenylmethylene)-

<b>Other names:</b>	Malonic acid, benzylidene- Benzylidenemalonic acid
<b>Inchi:</b>	InChI=1S/C10H8O4/c11-9(12)8(10(13)14)6-7-4-2-1-3-5-7/h1-6H,(H,11,12)(H,13,14)
<b>InchiKey:</b>	KXTAOXNYQGASTA-UHFFFAOYSA-N
<b>Formula:</b>	C10H8O4
<b>SMILES:</b>	O=C(O)C(=Cc1ccccc1)C(=O)O
<b>Mol. weight [g/mol]:</b>	192.17
<b>CAS:</b>	584-45-2

## Physical Properties

Property code	Value	Unit	Source
chs	-4363.10	kJ/mol	NIST Webbook
gf	-314.08	kJ/mol	Joback Method
hf	-435.39	kJ/mol	Joback Method
hfs	-715.00	kJ/mol	NIST Webbook
hfus	25.96	kJ/mol	Joback Method
hvap	87.02	kJ/mol	Joback Method
log10ws	-1.33		Crippen Method
logp	1.239		Crippen Method
mvol	138.580	ml/mol	McGowan Method
pc	4565.38	kPa	Joback Method
tb	751.02	K	Joback Method
tc	956.74	K	Joback Method
tf	431.34	K	Joback Method
vc	0.518	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	357.53	J/molxK	751.02	Joback Method
cpg	364.98	J/molxK	785.31	Joback Method
cpg	371.89	J/molxK	819.59	Joback Method
cpg	378.31	J/molxK	853.88	Joback Method
cpg	384.29	J/molxK	888.17	Joback Method

cpg	389.87	J/mol×K	922.45	Joback Method
cpg	395.09	J/mol×K	956.74	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C584452&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C584452&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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