

# 2-chlorobutyl trichloroacetate

<b>Other names:</b>	1-Butanol, 2-chloro, trichloroacetate
<b>Inchi:</b>	InChI=1S/C6H8Cl4O2/c1-2-4(7)3-12-5(11)6(8,9)10/h4H,2-3H2,1H3
<b>InchiKey:</b>	DUUQIODKAXWGHX-UHFFFAOYSA-N
<b>Formula:</b>	C6H8Cl4O2
<b>SMILES:</b>	CCC(Cl)COC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	253.94

## Physical Properties

Property code	Value	Unit	Source
gf	-281.60	kJ/mol	Joback Method
hf	-488.96	kJ/mol	Joback Method
hfus	19.93	kJ/mol	Joback Method
hvap	53.96	kJ/mol	Joback Method
log10ws	-3.02		Crippen Method
logp	2.917		Crippen Method
mcvol	151.800	ml/mol	McGowan Method
pc	2862.74	kPa	Joback Method
rinpol	1264.00		NIST Webbook
rinpol	1268.00		NIST Webbook
rinpol	1301.00		NIST Webbook
rinpol	1311.00		NIST Webbook
rinpol	1268.00		NIST Webbook
ripol	1846.00		NIST Webbook
ripol	1873.00		NIST Webbook
ripol	1859.00		NIST Webbook
ripol	1851.00		NIST Webbook
ripol	1828.00		NIST Webbook
tb	559.02	K	Joback Method
tc	776.12	K	Joback Method
tf	336.64	K	Joback Method
vc	0.575	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	298.23	J/mol×K	559.02	Joback Method
cpg	307.26	J/mol×K	595.20	Joback Method
cpg	315.65	J/mol×K	631.39	Joback Method
cpg	323.42	J/mol×K	667.57	Joback Method
cpg	330.62	J/mol×K	703.76	Joback Method
cpg	337.27	J/mol×K	739.94	Joback Method
cpg	343.39	J/mol×K	776.12	Joback Method
dvisc	0.0032699	Paxs	336.64	Joback Method
dvisc	0.0017065	Paxs	373.70	Joback Method
dvisc	0.0010015	Paxs	410.77	Joback Method
dvisc	0.0006420	Paxs	447.83	Joback Method
dvisc	0.0004404	Paxs	484.89	Joback Method
dvisc	0.0003188	Paxs	521.96	Joback Method
dvisc	0.0002408	Paxs	559.02	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R111567&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R111567&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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