

# Malonic acid, 8-chlorooctyl pentyl ester

<b>Inchi:</b>	InChI=1S/C16H29ClO4/c1-2-3-9-12-20-15(18)14-16(19)21-13-10-7-5-4-6-8-11-17/h2-14H
<b>InchiKey:</b>	GHRBRCOZQDGBSO-UHFFFAOYSA-N
<b>Formula:</b>	C16H29ClO4
<b>SMILES:</b>	CCCCCOC(=O)CC(=O)OCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	320.85

## Physical Properties

Property code	Value	Unit	Source
gf	-395.93	kJ/mol	Joback Method
hf	-878.91	kJ/mol	Joback Method
hfus	46.97	kJ/mol	Joback Method
hvap	73.91	kJ/mol	Joback Method
log10ws	-4.40		Crippen Method
logp	4.232		Crippen Method
mvol	263.420	ml/mol	McGowan Method
pc	1372.76	kPa	Joback Method
rinpol	2212.00		NIST Webbook
tb	755.49	K	Joback Method
tc	937.01	K	Joback Method
tf	444.32	K	Joback Method
vc	1.028	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	764.82	J/molxK	755.49	Joback Method
cpg	780.54	J/molxK	785.74	Joback Method
cpg	795.41	J/molxK	816.00	Joback Method
cpg	809.43	J/molxK	846.25	Joback Method
cpg	822.63	J/molxK	876.50	Joback Method
cpg	835.01	J/molxK	906.75	Joback Method
cpg	846.57	J/molxK	937.01	Joback Method
dvisc	0.0010176	Paxs	444.32	Joback Method
dvisc	0.0005360	Paxs	496.18	Joback Method

dvisc	0.0003188	Paxs	548.04	Joback Method
dvisc	0.0002074	Paxs	599.90	Joback Method
dvisc	0.0001445	Paxs	651.77	Joback Method
dvisc	0.0001062	Paxs	703.63	Joback Method
dvisc	0.0000814	Paxs	755.49	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U349006&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U349006&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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