

# Phenetole, o-iodo-

<b>Other names:</b>	Benzene, 1-ethoxy-2-iodo-
<b>Inchi:</b>	InChI=1S/C8H9IO/c1-2-10-8-6-4-3-5-7(8)9/h3-6H,2H2,1H3
<b>InchiKey:</b>	JUJREUXKABNEFA-UHFFFAOYSA-N
<b>Formula:</b>	C8H9IO
<b>SMILES:</b>	CCOc1cccc1I
<b>Mol. weight [g/mol]:</b>	248.06
<b>CAS:</b>	614-73-3

## Physical Properties

Property code	Value	Unit	Source
gf	72.38	kJ/mol	Joback Method
hf	-38.74	kJ/mol	Joback Method
hfus	15.72	kJ/mol	Joback Method
hvap	48.12	kJ/mol	Joback Method
log10ws	-3.16		Crippen Method
logp	2.690		Crippen Method
mcvol	131.510	ml/mol	McGowan Method
pc	3423.86	kPa	Joback Method
tb	529.66	K	Joback Method
tc	774.42	K	Joback Method
tf	299.15	K	Joback Method
vc	0.481	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	244.94	J/molxK	529.66	Joback Method
cpg	256.53	J/molxK	570.45	Joback Method
cpg	267.35	J/molxK	611.25	Joback Method
cpg	277.46	J/molxK	652.04	Joback Method
cpg	286.86	J/molxK	692.83	Joback Method
cpg	295.59	J/molxK	733.62	Joback Method
cpg	303.69	J/molxK	774.42	Joback Method
dvisc	0.0022252	Paxs	299.15	Joback Method

dvisc	0.0012464	Paxs	337.57	Joback Method
dvisc	0.0007860	Paxs	375.99	Joback Method
dvisc	0.0005398	Paxs	414.40	Joback Method
dvisc	0.0003952	Paxs	452.82	Joback Method
dvisc	0.0003038	Paxs	491.24	Joback Method
dvisc	0.0002426	Paxs	529.66	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	518.20	K	98.10	NIST Webbook
tbrp	399.20	K	2.40	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C614733&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C614733&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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