

Fumaric acid, 2-decyl ethyl ester

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|-----------------------------|--|
| Inchi: | InChI=1S/C16H28O4/c1-4-6-7-8-9-10-11-14(3)20-16(18)13-12-15(17)19-5-2/h12-14H,4- |
| InchiKey: | RMUGYXZVCWKTIZ-OUKQBFOZSA-N |
| Formula: | C16H28O4 |
| SMILES: | CCCCCCCCC(C)OC(=O)C=CC(=O)OCC |
| Mol. weight [g/mol]: | 284.39 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -306.22 | kJ/mol | Joback Method |
| hf | -751.23 | kJ/mol | Joback Method |
| hfus | 39.45 | kJ/mol | Joback Method |
| hvap | 69.09 | kJ/mol | Joback Method |
| log10ws | -4.21 | | Crippen Method |
| logp | 3.788 | | Crippen Method |
| mcvol | 246.880 | ml/mol | McGowan Method |
| pc | 1477.02 | kPa | Joback Method |
| rinsol | 1895.00 | | NIST Webbook |
| tb | 721.78 | K | Joback Method |
| tc | 904.51 | K | Joback Method |
| tf | 394.32 | K | Joback Method |
| vc | 0.954 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 710.90 | J/molxK | 721.78 | Joback Method |
| cpg | 783.99 | J/molxK | 874.05 | Joback Method |
| cpg | 770.99 | J/molxK | 843.60 | Joback Method |
| cpg | 757.19 | J/molxK | 813.14 | Joback Method |
| cpg | 742.59 | J/molxK | 782.69 | Joback Method |
| cpg | 727.17 | J/molxK | 752.23 | Joback Method |
| cpg | 796.23 | J/molxK | 904.51 | Joback Method |
| dvisc | 0.0000739 | Paxs | 721.78 | Joback Method |
| dvisc | 0.0000989 | Paxs | 667.20 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0001394 | Paxs | 612.63 | Joback Method |
| dvisc | 0.0002100 | Paxs | 558.05 | Joback Method |
| dvisc | 0.0003457 | Paxs | 503.47 | Joback Method |
| dvisc | 0.0006427 | Paxs | 448.90 | Joback Method |
| dvisc | 0.0014186 | Paxs | 394.32 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=U348584&Units=SI |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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<https://www.chemeo.com/cid/54-454-8/Fumaric-acid-2-decyl-ethyl-ester.pdf>

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