

# Butane, 2-methoxy-2,3,3-trimethyl-

<b>Other names:</b>	Ether, methyl 1,1,2,2-tetramethylpropyl 2-Methoxy-2,3,3-trimethylbutane
<b>Inchi:</b>	InChI=1S/C8H18O/c1-7(2,3)8(4,5)9-6/h1-6H3
<b>InchiKey:</b>	FTACTAWTKDTQSA-UHFFFAOYSA-N
<b>Formula:</b>	C8H18O
<b>SMILES:</b>	COC(C)(C)C(C)(C)C
<b>Mol. weight [g/mol]:</b>	130.23
<b>CAS:</b>	27705-21-1

## Physical Properties

Property code	Value	Unit	Source
gf	-82.84	kJ/mol	Joback Method
hf	-358.17	kJ/mol	Joback Method
hfus	2.84	kJ/mol	Joback Method
hvap	33.22	kJ/mol	Joback Method
log10ws	-2.13		Crippen Method
logp	2.458		Crippen Method
mcvol	129.450	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
tb	398.40	K	Joback Method
tc	584.50	K	Joback Method
tf	206.99	K	Joback Method
vc	0.479	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	260.30	J/mol×K	398.40	Joback Method
cpg	276.06	J/mol×K	429.42	Joback Method
cpg	291.01	J/mol×K	460.43	Joback Method
cpg	305.18	J/mol×K	491.45	Joback Method
cpg	318.61	J/mol×K	522.47	Joback Method
cpg	331.31	J/mol×K	553.49	Joback Method
cpg	343.33	J/mol×K	584.50	Joback Method

dvisc	0.0127059	Paxs	206.99	Joback Method
dvisc	0.0043347	Paxs	238.89	Joback Method
dvisc	0.0019053	Paxs	270.79	Joback Method
dvisc	0.0009959	Paxs	302.69	Joback Method
dvisc	0.0005891	Paxs	334.60	Joback Method
dvisc	0.0003818	Paxs	366.50	Joback Method
dvisc	0.0002653	Paxs	398.40	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C27705211&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C27705211&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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