

# Diethylmalonic acid, 3-methylphenyl octyl ester

Inchi:	InChI=1S/C22H34O4/c1-5-8-9-10-11-12-16-25-20(23)22(6-2,7-3)21(24)26-19-15-13-14-1
InchiKey:	NZGNZXPWMMHZNI-UHFFFAOYSA-N
Formula:	C22H34O4
SMILES:	CCCCCCCCOC(=O)C(CC)(CC)C(=O)Oc1cccc(C)c1
Mol. weight [g/mol]:	362.50

## Physical Properties

Property code	Value	Unit	Source
gf	-227.86	kJ/mol	Joback Method
hf	-770.70	kJ/mol	Joback Method
hfus	44.55	kJ/mol	Joback Method
hvap	84.52	kJ/mol	Joback Method
log10ws	-6.32		Crippen Method
logp	5.611		Crippen Method
mcvol	311.960	ml/mol	McGowan Method
pc	1186.60	kPa	Joback Method
rinsol	2336.00		NIST Webbook
tb	883.77	K	Joback Method
tc	1090.17	K	Joback Method
tf	523.38	K	Joback Method
vc	1.196	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	997.84	J/molxK	883.77	Joback Method
cpg	1014.47	J/molxK	918.17	Joback Method
cpg	1029.90	J/molxK	952.57	Joback Method
cpg	1044.16	J/molxK	986.97	Joback Method
cpg	1057.31	J/molxK	1021.37	Joback Method
cpg	1069.39	J/molxK	1055.77	Joback Method
cpg	1080.45	J/molxK	1090.17	Joback Method
dvisc	0.0004346	Paxs	523.38	Joback Method
dvisc	0.0002253	Paxs	583.44	Joback Method

dvisc	0.0001320	Paxs	643.51	Joback Method
dvisc	0.0000847	Paxs	703.57	Joback Method
dvisc	0.0000583	Paxs	763.64	Joback Method
dvisc	0.0000424	Paxs	823.70	Joback Method
dvisc	0.0000322	Paxs	883.77	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370016&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370016&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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