

# Propanoic acid, 3-bromo-2-oxo-, ethyl ester

<b>Other names:</b>	Pyruvic acid, bromo-, ethyl ester «beta»-Bromopyruvic acid ethyl ester Ethyl bromopyruvate Bromopyruvic acid ethyl ester Ethyl 3-bromopyruvate
<b>Inchi:</b>	InChI=1S/C5H7BrO3/c1-2-9-5(8)4(7)3-6/h2-3H2,1H3
<b>InchiKey:</b>	VICYTAYPKBLQFB-UHFFFAOYSA-N
<b>Formula:</b>	C5H7BrO3
<b>SMILES:</b>	CCOC(=O)C(=O)CBr
<b>Mol. weight [g/mol]:</b>	195.01
<b>CAS:</b>	70-23-5

## Physical Properties

Property code	Value	Unit	Source
gf	-357.30	kJ/mol	Joback Method
hf	-477.58	kJ/mol	Joback Method
hfus	18.38	kJ/mol	Joback Method
hvap	49.06	kJ/mol	Joback Method
log10ws	-0.49		Crippen Method
logp	0.513		Crippen Method
mcvol	107.820	ml/mol	McGowan Method
pc	4328.25	kPa	Joback Method
tb	510.12	K	Joback Method
tc	716.96	K	Joback Method
tf	328.00	K	Joback Method
vc	0.407	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	206.40	J/molxK	510.12	Joback Method
cpg	241.54	J/molxK	682.49	Joback Method
cpg	235.24	J/molxK	648.01	Joback Method
cpg	228.58	J/molxK	613.54	Joback Method

cpg	221.56	J/molxK	579.07	Joback Method
cpg	214.16	J/molxK	544.59	Joback Method
cpg	247.48	J/molxK	716.96	Joback Method
dvisc	0.0003799	Paxs	510.12	Joback Method
dvisc	0.0004679	Paxs	479.77	Joback Method
dvisc	0.0005928	Paxs	449.41	Joback Method
dvisc	0.0007772	Paxs	419.06	Joback Method
dvisc	0.0010630	Paxs	388.71	Joback Method
dvisc	0.0015331	Paxs	358.35	Joback Method
dvisc	0.0023661	Paxs	328.00	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	372.20	K	1.30	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C70235&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C70235&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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