

# Thioxanthene

<b>Other names:</b>	9H-Thioxanthene Dibenzothiapyran Thiaxanthen Thiaxanthene Thioxanthen 10H-dibenzo[b,e]thiin
<b>Inchi:</b>	InChI=1S/C13H10S/c1-3-7-12-10(5-1)9-11-6-2-4-8-13(11)14-12/h1-8H,9H2
<b>InchiKey:</b>	PQJUJGAVDBINPI-UHFFFAOYSA-N
<b>Formula:</b>	C13H10S
<b>SMILES:</b>	<chem>c1ccc2c(c1)Cc1ccccc1S2</chem>
<b>Mol. weight [g/mol]:</b>	198.28
<b>CAS:</b>	261-31-4

## Physical Properties

Property code	Value	Unit	Source
gf	384.56	kJ/mol	Joback Method
hf	283.03	kJ/mol	Joback Method
hfus	19.55	kJ/mol	Joback Method
hsub	101.70 ± 1.60	kJ/mol	NIST Webbook
hsub	100.90 ± 0.20	kJ/mol	NIST Webbook
hvap	77.80 ± 2.60	kJ/mol	NIST Webbook
log10ws	-4.23		Crippen Method
logp	3.742		Crippen Method
mcvol	152.000	ml/mol	McGowan Method
pc	3452.08	kPa	Joback Method
tb	615.13	K	Joback Method
tc	884.01	K	Joback Method
tf	401.70 ± 0.10	K	NIST Webbook
vc	0.559	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	350.25	J/mol×K	615.13	Joback Method

cpg	364.96	J/mol×K	659.94	Joback Method
cpg	378.32	J/mol×K	704.76	Joback Method
cpg	390.49	J/mol×K	749.57	Joback Method
cpg	401.65	J/mol×K	794.38	Joback Method
cpg	411.95	J/mol×K	839.20	Joback Method
cpg	421.57	J/mol×K	884.01	Joback Method
hfust	26.10	kJ/mol	401.80	NIST Webbook
hsubt	98.40 ± 0.20	kJ/mol	370.50	NIST Webbook
hvapt	69.50 ± 0.20	kJ/mol	415.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C261314&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C261314&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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