

# Benzoic acid, 3,5-dinitro, 1-phenylethyl ester

**Inchi:** InChI=1S/C15H12N2O6/c1-10(11-5-3-2-4-6-11)23-15(18)12-7-13(16(19)20)9-14(8-12)17  
**InchiKey:** YATKDADIXNBBCZ-UHFFFAOYSA-N  
**Formula:** C15H12N2O6  
**SMILES:** CC(OC(=O)c1cc([N+](=O)[O-])cc([N+](=O)[O-])c1)cccc1  
**Mol. weight [g/mol]:** 316.27

## Physical Properties

Property code	Value	Unit	Source
gf	115.72	kJ/mol	Joback Method
hf	-174.41	kJ/mol	Joback Method
hfus	43.90	kJ/mol	Joback Method
hvap	96.81	kJ/mol	Joback Method
log10ws	-5.51		Crippen Method
logp	3.421		Crippen Method
mcvol	216.970	ml/mol	McGowan Method
pc	2635.25	kPa	Joback Method
rinpol	2344.00		NIST Webbook
rinpol	2327.00		NIST Webbook
rinpol	2310.00		NIST Webbook
rinpol	2370.00		NIST Webbook
rinpol	2362.00		NIST Webbook
rinpol	2377.00		NIST Webbook
rinpol	2358.00		NIST Webbook
rinpol	2362.00		NIST Webbook
tb	985.45	K	Joback Method
tc	1260.43	K	Joback Method
tf	681.07	K	Joback Method
vc	0.842	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	647.67	J/molxK	985.45	Joback Method
cpg	656.37	J/molxK	1031.28	Joback Method

cpg	663.79	J/mol×K	1077.11	Joback Method
cpg	670.00	J/mol×K	1122.94	Joback Method
cpg	675.09	J/mol×K	1168.77	Joback Method
cpg	679.15	J/mol×K	1214.60	Joback Method
cpg	682.25	J/mol×K	1260.43	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R35068&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R35068&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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