

«beta»-Alanine, N-(2,6-difluorobenzoyl)-, undecyl ester

Inchi:	InChI=1S/C21H31F2NO3/c1-2-3-4-5-6-7-8-9-10-16-27-19(25)14-15-24-21(26)20-17(22)1
InchiKey:	ULDHCRXNENUWRS-UHFFFAOYSA-N
Formula:	C21H31F2NO3
SMILES:	CCCCCCCCCOC(=O)CCNC(=O)c1c(F)cccc1F
Mol. weight [g/mol]:	383.47

Physical Properties

Property code	Value	Unit	Source
gf	-443.98	kJ/mol	Joback Method
hf	-959.31	kJ/mol	Joback Method
hfus	59.05	kJ/mol	Joback Method
hvap	86.64	kJ/mol	Joback Method
log10ws	-6.78		Crippen Method
logp	5.159		Crippen Method
mcvol	305.520	ml/mol	McGowan Method
pc	1188.24	kPa	Joback Method
rinpol	2807.00		NIST Webbook
rinpol	2807.00		NIST Webbook
tb	895.39	K	Joback Method
tc	1097.39	K	Joback Method
tf	553.82	K	Joback Method
vc	1.204	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	981.39	J/mol×K	895.39	Joback Method
cpg	996.82	J/mol×K	929.06	Joback Method
cpg	1011.12	J/mol×K	962.72	Joback Method
cpg	1024.35	J/mol×K	996.39	Joback Method
cpg	1036.52	J/mol×K	1030.05	Joback Method
cpg	1047.69	J/mol×K	1063.72	Joback Method
cpg	1057.90	J/mol×K	1097.39	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U321849&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvp:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinp:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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