

# Benzene, 1,2-dimethyl-3-nitro-

<b>Other names:</b>	1,2-Dimethyl-3-nitrobenzene 2,3-Dimethylnitrobenzene 3-Nitro-o-xylene o-Xylene, 3-nitro-
<b>Inchi:</b>	InChI=1S/C8H9NO2/c1-6-4-3-5-8(7(6)2)9(10)11/h3-5H,1-2H3
<b>InchiKey:</b>	FVHAWXWFPBPFOS-UHFFFAOYSA-N
<b>Formula:</b>	C8H9NO2
<b>SMILES:</b>	<chem>Cc1cccc([N+](=O)[O-])c1C</chem>
<b>Mol. weight [g/mol]:</b>	151.16
<b>CAS:</b>	83-41-0

## Physical Properties

Property code	Value	Unit	Source
ea	0.92 ± 0.05	eV	NIST Webbook
ea	0.85 ± 0.10	eV	NIST Webbook
gf	145.18	kJ/mol	Joback Method
hf	-5.62	kJ/mol	Joback Method
hfus	21.10	kJ/mol	Joback Method
hvap	53.59	kJ/mol	Joback Method
log10ws	-3.07		Crippen Method
logp	2.212		Crippen Method
mvol	117.240	ml/mol	McGowan Method
pc	3572.80	kPa	Joback Method
tb	518.20	K	NIST Webbook
tc	816.66	K	Joback Method
tf	287.40 ± 1.00	K	NIST Webbook
vc	0.458	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.17	J/mol×K	775.71	Joback Method
cpg	264.31	J/mol×K	570.92	Joback Method
cpg	276.23	J/mol×K	611.88	Joback Method

cpg	287.35	J/mol×K	652.83	Joback Method
cpg	297.69	J/mol×K	693.79	Joback Method
cpg	307.28	J/mol×K	734.75	Joback Method
cpg	324.37	J/mol×K	816.66	Joback Method
hvapt	59.40	kJ/mol	450.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.45741e+01
Coeff. B	-4.32473e+03
Coeff. C	-8.38060e+01
Temperature range (K), min.	386.52
Temperature range (K), max.	550.71

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C83410&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C83410&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/54-860-7/Benzene-1-2-dimethyl-3-nitro.pdf>

Generated by Cheméo on 2024-04-20 04:01:17.012241844 +0000 UTC m=+15874925.932819160.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.