

# 1,3-Dimethyl-2,4,5,6-tetrakis(chloromethyl)benzene

<b>Inchi:</b>	InChI=1S/C12H14Cl4/c1-7-9(3-13)8(2)11(5-15)12(6-16)10(7)4-14/h3-6H2,1-2H3
<b>InchiKey:</b>	MKOKIPCCVXZHAR-UHFFFAOYSA-N
<b>Formula:</b>	C12H14Cl4
<b>SMILES:</b>	Cc1c(CCl)c(C)c(CCl)c(CCl)c1CCl
<b>Mol. weight [g/mol]:</b>	300.05

## Physical Properties

Property code	Value	Unit	Source
gf	66.70	kJ/mol	Joback Method
hf	-174.79	kJ/mol	Joback Method
hfus	35.72	kJ/mol	Joback Method
hvap	65.43	kJ/mol	Joback Method
log10ws	-6.55		Crippen Method
logp	5.259		Crippen Method
mcvol	205.140	ml/mol	McGowan Method
pc	1954.41	kPa	Joback Method
rinsol	2152.00		NIST Webbook
tb	675.26	K	Joback Method
tc	895.74	K	Joback Method
tf	433.70	K	Joback Method
vc	0.795	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	453.44	J/molxK	675.26	Joback Method
cpg	465.56	J/molxK	712.01	Joback Method
cpg	476.97	J/molxK	748.75	Joback Method
cpg	487.70	J/molxK	785.50	Joback Method
cpg	497.76	J/molxK	822.25	Joback Method
cpg	507.19	J/molxK	859.00	Joback Method
cpg	515.99	J/molxK	895.74	Joback Method
dvisc	0.0008017	Paxs	433.70	Joback Method
dvisc	0.0005523	Paxs	473.96	Joback Method

dvisc	0.0004033	Paxs	514.22	Joback Method
dvisc	0.0003083	Paxs	554.48	Joback Method
dvisc	0.0002444	Paxs	594.74	Joback Method
dvisc	0.0001995	Paxs	635.00	Joback Method
dvisc	0.0001669	Paxs	675.26	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R520293&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R520293&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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