

# Phenol, 2-methyl, 4,6-dinitro-

**Other names:** 2,4-Dinitro-6-methylphenol  
2,4-Dinitro-o-cresol  
2-Methyl-4,6-dinitrophenol  
3,5-Dinitro-2-hydroxytoluene  
4,6-DNOC  
4,6-Dinitro-2-methylphenol  
4,6-Dinitro-o-cresol  
4,6-Dinitro-o-cresolo  
4,6-Dinitro-o-kresol  
4,6-Dinitrocresol  
4,6-Dinitrokresol  
6-Methyl-2,4-dinitrophenol  
Antinonin  
Antinonnin  
Arborol  
C.I. 10310  
Capsine  
Chemsect DNOC  
DINOK  
DNC  
Degrassan  
Dekrysil  
Detal  
Detol  
Dillex  
Dinitro  
Dinitro-o-cresol  
Dinitrocresol  
Dinitrodendtroxal  
Dinitrol  
Dinitrosol  
Dinoc  
Dinurania  
Ditrosol  
Dn  
Dn-dry mix no. 2  
Dnoc  
Dnok  
Dwunitro-o-krezol  
ENT 154

Effusan  
Effusan 3436  
Elgetol  
Elgetol 30  
Elgetox  
Elipol  
Extrar  
Flavin-Sandoz  
Hedolit  
Hedolite  
K III  
K IV  
Krenite  
Kresamone  
Kresonite-E  
Krezotol 50  
Le dinitrocresol-4,6  
Lipan  
Nitrador  
Nitrofan  
Oranz viktoria  
Prokarbol  
RCRA Waste number P047  
Rafex  
Rafex 35  
Raphatox  
Sandolin  
Sandolin A  
Selinon  
Sinox  
Trifocide  
Trifrina  
Winterwash  
Zahlreiche bezeichnungen  
o-Cresol, 4,6-dinitro-

**Inchi:** InChI=1S/C7H6N2O5/c1-4-2-5(8(11)12)3-6(7(4)10)9(13)14/h2-3,10H,1H3  
**InchiKey:** ZXVONLUNISGICL-UHFFFAOYSA-N  
**Formula:** C7H6N2O5  
**SMILES:** Cc1cc([N+](=O)[O-])cc([N+](=O)[O-])c1O  
**Mol. weight [g/mol]:** 198.13  
**CAS:** 534-52-1

# Physical Properties

Property code	Value	Unit	Source
chs	-3332.90	kJ/mol	NIST Webbook
gf	17.69	kJ/mol	Joback Method
hf	-173.05	kJ/mol	Joback Method
hfs	-279.00	kJ/mol	NIST Webbook
hfus	35.65	kJ/mol	Joback Method
hvap	80.97	kJ/mol	Joback Method
log10ws	-1.46		Estimated Solubility Method
log10ws	-1.46		Aqueous Solubility Prediction Method
logp	1.517		Crippen Method
mcvol	126.440	ml/mol	McGowan Method
pc	4987.39	kPa	Joback Method
rinpol	274.10		NIST Webbook
rinpol	274.10		NIST Webbook
rinpol	1564.00		NIST Webbook
rinpol	1634.60		NIST Webbook
rinpol	1634.60		NIST Webbook
rinpol	1624.00		NIST Webbook
rinpol	1607.00		NIST Webbook
rinpol	1617.00		NIST Webbook
rinpol	1617.00		NIST Webbook
rinpol	1564.00		NIST Webbook
rinpol	275.47		NIST Webbook
tb	780.50	K	Joback Method
tc	1059.76	K	Joback Method
tf	359.75	K	Aqueous Solubility Prediction Method
tf	360.25 ± 0.20	K	NIST Webbook
tf	359.00 ± 0.20	K	NIST Webbook
vc	0.450	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.95	J/mol×K	780.50	Joback Method
cpg	343.28	J/mol×K	827.04	Joback Method

cpg	351.09	J/mol×K	873.59	Joback Method
cpg	358.52	J/mol×K	920.13	Joback Method
cpg	365.69	J/mol×K	966.67	Joback Method
cpg	372.74	J/mol×K	1013.21	Joback Method
cpg	379.80	J/mol×K	1059.76	Joback Method
hfust	19.41	kJ/mol	359.30	NIST Webbook
hfust	19.41	kJ/mol	359.30	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C534521&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C534521&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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