

# 4,4-Dimethyl-1-hexene

<b>Other names:</b>	1-Hexene, 4,4-dimethyl- 4,4-Dimethylhex-1-ene
<b>Inchi:</b>	InChI=1S/C8H16/c1-5-7-8(3,4)6-2/h5H,1,6-7H2,2-4H3
<b>InchiKey:</b>	SUJVAMIXNUAJEY-UHFFFAOYSA-N
<b>Formula:</b>	C8H16
<b>SMILES:</b>	C=CCC(C)(C)CC
<b>Mol. weight [g/mol]:</b>	112.21
<b>CAS:</b>	1647-08-1

## Physical Properties

Property code	Value	Unit	Source
gf	107.16	kJ/mol	Joback Method
hf	-91.77	kJ/mol	Joback Method
hfus	7.78	kJ/mol	Joback Method
hvap	31.00	kJ/mol	NIST Webbook
log10ws	-2.78		Crippen Method
logp	2.999		Crippen Method
mcvol	119.280	ml/mol	McGowan Method
pc	2698.60	kPa	Joback Method
rinpol	728.10		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	725.00		NIST Webbook
rinpol	723.30		NIST Webbook
tb	375.89	K	Joback Method
tc	553.76	K	Joback Method
tf	180.58	K	Joback Method
vc	0.454	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	218.61	J/molxK	375.89	Joback Method
cpg	232.71	J/molxK	405.54	Joback Method
cpg	246.11	J/molxK	435.18	Joback Method
cpg	258.84	J/molxK	464.83	Joback Method
cpg	270.93	J/molxK	494.47	Joback Method
cpg	282.41	J/molxK	524.12	Joback Method
cpg	293.30	J/molxK	553.76	Joback Method
dvisc	0.0095731	Paxs	180.58	Joback Method
dvisc	0.0033567	Paxs	213.13	Joback Method
dvisc	0.0015537	Paxs	245.68	Joback Method
dvisc	0.0008612	Paxs	278.24	Joback Method
dvisc	0.0005402	Paxs	310.79	Joback Method
dvisc	0.0003702	Paxs	343.34	Joback Method
dvisc	0.0002708	Paxs	375.89	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.26501e+01
Coeff. B	-2.64563e+03
Coeff. C	-6.39590e+01
Temperature range (K), min.	277.96
Temperature range (K), max.	424.47

## Sources

**The Yaws Handbook of Vapor Pressure:**  
**Crippen Method:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**KDB:**

<https://www.thermo.com/files/research/kdb/mol/mol289.mol>

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1647081&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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