

# Procaine

## Other names:

.beta.-(diethylamino)ethyl p-aminobenzoate  
.beta.-diethylaminoethyl 4-aminobenzoate  
2-(Diethylamino)ethyl 4-aminobenzoate  
2-(Diethylamino)ethyl p-aminobenzoate  
2-Diethylaminoethyl 4-aminobenzoate  
2-Diethylaminoethylester kyseliny p-aminobenzoove  
4-Aminobenzoic acid diethylaminoethyl ester  
Benzoic acid, 4-amino-, 2-(diethylamino)ethyl ester  
Benzoic acid, p-amino-, 2-(diethylamino)ethyl ester  
Diethylaminoethyl p-aminobenzoate  
Duracaine  
NSC 169497  
Nissocaine  
Procain  
Spinocaine  
Vitamin H3  
benzoic acid, p-amino-, 2-(diethylamino) ethyl ester  
p-Aminobenzoic acid, 2-(diethylamino)ethyl ester  
p-Aminobenzoyldiethylaminoethanol  
p-aminobenzoic acid 2-diethylaminoethyl ester  
procaine base  
«beta»-(Diethylamino)ethyl 4-aminobenzoate  
«beta»-(Diethylamino)ethyl p-aminobenzoate  
Â«betaÂ»-(Diethylamino)ethyl 4-aminobenzoate  
Â«betaÂ»-(Diethylamino)ethyl p-aminobenzoate

**Inchi:** InChI=1S/C13H20N2O2/c1-3-15(4-2)9-10-17-13(16)11-5-7-12(14)8-6-11/h5-8H,3-4,9-10

**InchiKey:** MFDFERRIHVXMIY-UHFFFAOYSA-N

**Formula:** C13H20N2O2

**SMILES:** CCN(CC)CCOC(=O)c1ccc(N)cc1

**Mol. weight [g/mol]:** 236.31

**CAS:** 59-46-1

## Physical Properties

Property code	Value	Unit	Source
gf	104.67	kJ/mol	Joback Method
hf	-230.07	kJ/mol	Joback Method

hfus	34.08		kJ/mol	Joback Method
hvap	69.31		kJ/mol	Joback Method
log10ws	-1.50			Aqueous Solubility Prediction Method
logp	1.767			Crippen Method
mcvol	197.670		ml/mol	McGowan Method
pc	2381.86		kPa	Joback Method
rinpol	2029.00			NIST Webbook
rinpol	2031.00			NIST Webbook
rinpol	2046.00			NIST Webbook
rinpol	2049.00			NIST Webbook
rinpol	2014.00			NIST Webbook
rinpol	2035.00			NIST Webbook
rinpol	1998.00			NIST Webbook
rinpol	2005.00			NIST Webbook
rinpol	2007.00			NIST Webbook
rinpol	2000.00			NIST Webbook
rinpol	1995.00			NIST Webbook
rinpol	2013.00			NIST Webbook
rinpol	1984.00			NIST Webbook
rinpol	1990.00			NIST Webbook
rinpol	2058.00			NIST Webbook
rinpol	1978.00			NIST Webbook
rinpol	2000.00			NIST Webbook
rinpol	1991.00			NIST Webbook
rinpol	1995.00			NIST Webbook
rinpol	2007.00			NIST Webbook
rinpol	2004.00			NIST Webbook
rinpol	2027.00			NIST Webbook
rinpol	2018.00			NIST Webbook
rinpol	2010.00			NIST Webbook
rinpol	1978.00			NIST Webbook
rinpol	1990.00			NIST Webbook
rinpol	2025.00			NIST Webbook
rinpol	1979.00			NIST Webbook
rinpol	2017.00			NIST Webbook
rinpol	1991.00			NIST Webbook
rinpol	1995.00			NIST Webbook
rinpol	2018.00			NIST Webbook
rinpol	2010.00			NIST Webbook
rinpol	1995.00			NIST Webbook
rinpol	2014.00			NIST Webbook
rinpol	2035.00			NIST Webbook
rinpol	2000.00			NIST Webbook
rinpol	2000.00			NIST Webbook

rmpol	2018.00		NIST Webbook
rmpol	2018.00		NIST Webbook
ripol	3250.00		NIST Webbook
ripol	3250.00		NIST Webbook
tb	689.76	K	Joback Method
tc	898.32	K	Joback Method
tf	463.10	K	Joback Method
vc	0.727	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	546.04	J/mol×K	689.76	Joback Method
cpg	561.16	J/mol×K	724.52	Joback Method
cpg	575.33	J/mol×K	759.28	Joback Method
cpg	588.58	J/mol×K	794.04	Joback Method
cpg	600.94	J/mol×K	828.80	Joback Method
cpg	612.45	J/mol×K	863.56	Joback Method
cpg	623.13	J/mol×K	898.32	Joback Method

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C59461&Units=SI>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

The Solubility of Benzocaine, Lidocaine, and Procaine in Liquid and Supercritical CO<sub>2</sub>: The Solubility of Procaine HCl in Some Ethanol + Water Co-solvent Mixtures:

<https://www.doi.org/10.1021/je034163p>

<https://www.doi.org/10.1021/je900958z>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

Aqueous Solubility Prediction Method:

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

## Legend

**cpg:** Ideal gas heat capacity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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