

# Pentanedioic acid, 3,3-dimethyl-, dimethyl ester

<b>Other names:</b>	Glutaric acid, 3,3-dimethyl-, dimethyl ester Methyl 3,3-dimethylglutarate
<b>Inchi:</b>	InChI=1S/C9H16O4/c1-9(2,5-7(10)12-3)6-8(11)13-4/h5-6H2,1-4H3
<b>InchiKey:</b>	NPEAPWIQJBAJIO-UHFFFAOYSA-N
<b>Formula:</b>	C9H16O4
<b>SMILES:</b>	COC(=O)CC(C)(C)CC(=O)OC
<b>Mol. weight [g/mol]:</b>	188.22
<b>CAS:</b>	19184-67-9

## Physical Properties

Property code	Value	Unit	Source
gf	-440.10	kJ/mol	Joback Method
hf	-727.44	kJ/mol	Joback Method
hfus	17.23	kJ/mol	Joback Method
hvap	52.64	kJ/mol	Joback Method
log10ws	-1.07		Crippen Method
logp	1.139		Crippen Method
mvol	152.550	ml/mol	McGowan Method
pc	2574.11	kPa	Joback Method
tb	554.67	K	Joback Method
tc	746.71	K	Joback Method
tf	337.93	K	Joback Method
vc	0.577	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	371.44	J/molxK	554.67	Joback Method
cpg	384.56	J/molxK	586.68	Joback Method
cpg	397.05	J/molxK	618.68	Joback Method
cpg	408.91	J/molxK	650.69	Joback Method
cpg	420.16	J/molxK	682.69	Joback Method
cpg	430.80	J/molxK	714.70	Joback Method
cpg	440.85	J/molxK	746.71	Joback Method

dvisc	0.0022755	Paxs	337.93	Joback Method
dvisc	0.0012362	Paxs	374.05	Joback Method
dvisc	0.0007478	Paxs	410.18	Joback Method
dvisc	0.0004907	Paxs	446.30	Joback Method
dvisc	0.0003430	Paxs	482.42	Joback Method
dvisc	0.0002520	Paxs	518.55	Joback Method
dvisc	0.0001927	Paxs	554.67	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	376.70	K	2.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C19184679&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C19184679&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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