

Cyclohexane, 1,1'-tetradecylidenebis-

Other names:	1,1-Dicyclohexyltetradecane
Inchi:	InChI=1S/C26H50/c1-2-3-4-5-6-7-8-9-10-11-18-23-26(24-19-14-12-15-20-24)25-21-16-1
InchiKey:	COJYJGCWESKHMR-UHFFFAOYSA-N
Formula:	C26H50
SMILES:	CCCCCCCCCCCC(C1CCCCC1)C1CCCCC1
Mol. weight [g/mol]:	362.68
CAS:	55334-08-2

Physical Properties

Property code	Value	Unit	Source
gf	214.50	kJ/mol	Joback Method
hf	-476.61	kJ/mol	Joback Method
hfus	43.24	kJ/mol	Joback Method
hvap	73.94	kJ/mol	Joback Method
log10ws	-9.77		Crippen Method
logp	9.464		Crippen Method
mcvol	355.480	ml/mol	McGowan Method
pc	920.50	kPa	Joback Method
tb	832.94	K	Joback Method
tc	1032.60	K	Joback Method
tf	310.80 ± 4.00	K	NIST Webbook
vc	1.351	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1218.96	J/mol×K	832.94	Joback Method
cpg	1244.47	J/mol×K	866.22	Joback Method
cpg	1268.36	J/mol×K	899.49	Joback Method
cpg	1290.70	J/mol×K	932.77	Joback Method
cpg	1311.55	J/mol×K	966.04	Joback Method
cpg	1331.00	J/mol×K	999.32	Joback Method
cpg	1349.10	J/mol×K	1032.60	Joback Method
dvisc	0.0031936	Paxs	382.54	Joback Method

dvisc	0.0008654	Paxs	457.61	Joback Method
dvisc	0.0003388	Paxs	532.67	Joback Method
dvisc	0.0001672	Paxs	607.74	Joback Method
dvisc	0.0000964	Paxs	682.81	Joback Method
dvisc	0.0000620	Paxs	757.87	Joback Method
dvisc	0.0000432	Paxs	832.94	Joback Method
hvapt	97.70	kJ/mol	511.00	NIST Webbook

Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.28499e+01
Coeff. B	-4.75314e+03
Coeff. C	-1.29954e+02
Temperature range (K), min.	508.32
Temperature range (K), max.	760.48

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C55334082&Units=SI
The Yaws Handbook of Vapor Pressure:	https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
dvisc:	Dynamic viscosity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions

h_{vap}:	Enthalpy of vaporization at standard conditions
h_{vapt}:	Enthalpy of vaporization at a given temperature
log₁₀w_s:	Log10 of Water solubility in mol/l
log_p:	Octanol/Water partition coefficient
m_{cvol}:	McGowan's characteristic volume
p_c:	Critical Pressure
p_{vap}:	Vapor pressure
t_b:	Normal Boiling Point Temperature
t_c:	Critical Temperature
t_f:	Normal melting (fusion) point
v_c:	Critical Volume

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