

# 2-acetyl-2-methylthiazolidine

<b>Inchi:</b>	InChI=1S/C6H11NOS/c1-5(8)6(2)7-3-4-9-6/h7H,3-4H2,1-2H3
<b>InchiKey:</b>	WLXIBDDBHZVOIL-UHFFFAOYSA-N
<b>Formula:</b>	C6H11NOS
<b>SMILES:</b>	CC(=O)C1(C)NCCS1
<b>Mol. weight [g/mol]:</b>	145.22

## Physical Properties

Property code	Value	Unit	Source
gf	29.35	kJ/mol	Joback Method
hf	-120.96	kJ/mol	Joback Method
hfus	13.78	kJ/mol	Joback Method
hvap	47.37	kJ/mol	Joback Method
log10ws	-1.19		Crippen Method
logp	0.628		Crippen Method
mcvol	112.440	ml/mol	McGowan Method
pc	4522.49	kPa	Joback Method
rinpol	1134.00		NIST Webbook
rinpol	1134.00		NIST Webbook
rinpol	1155.00		NIST Webbook
rinpol	1155.00		NIST Webbook
ripol	1789.00		NIST Webbook
ripol	1789.00		NIST Webbook
ripol	1789.00		NIST Webbook
tb	502.45	K	Joback Method
tc	744.89	K	Joback Method
tf	430.59	K	Joback Method
vc	0.400	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	239.19	J/mol×K	502.45	Joback Method
cpg	252.38	J/mol×K	542.86	Joback Method
cpg	264.57	J/mol×K	583.26	Joback Method

cpg	275.89	J/mol×K	623.67	Joback Method
cpg	286.50	J/mol×K	664.08	Joback Method
cpg	296.53	J/mol×K	704.49	Joback Method
cpg	306.13	J/mol×K	744.89	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R187712&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R187712&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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