

# 3-Buten-2-one, 4-(4-iodophenyl)-

<b>Other names:</b>	(E)-4-(4-Iodophenyl)but-3-en-2-one
<b>Inchi:</b>	InChI=1S/C10H9IO/c1-8(12)2-3-9-4-6-10(11)7-5-9/h2-7H,1H3/b3-2+
<b>InchiKey:</b>	RNZRSAAJEHQBKM-NSCUHMNNSA-N
<b>Formula:</b>	C10H9IO
<b>SMILES:</b>	CC(=O)C=Cc1ccc(I)cc1
<b>Mol. weight [g/mol]:</b>	272.08
<b>CAS:</b>	18175-21-8

## Physical Properties

Property code	Value	Unit	Source
gf	145.52	kJ/mol	Joback Method
hf	56.84	kJ/mol	Joback Method
hfus	21.52	kJ/mol	Joback Method
hvap	56.87	kJ/mol	Joback Method
ie	8.40 ± 0.05	eV	NIST Webbook
log10ws	-3.56		Crippen Method
logp	2.893		Crippen Method
mcvol	151.090	ml/mol	McGowan Method
pc	3195.54	kPa	Joback Method
tb	611.03	K	Joback Method
tc	868.38	K	Joback Method
tf	344.31	K	Joback Method
vc	0.561	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	303.29	J/mol×K	611.03	Joback Method
cpg	314.98	J/mol×K	653.92	Joback Method
cpg	325.70	J/mol×K	696.81	Joback Method
cpg	335.53	J/mol×K	739.71	Joback Method
cpg	344.56	J/mol×K	782.60	Joback Method
cpg	352.88	J/mol×K	825.49	Joback Method
cpg	360.58	J/mol×K	868.38	Joback Method

dvisc	0.0022563	Paxs	344.31	Joback Method
dvisc	0.0012361	Paxs	388.76	Joback Method
dvisc	0.0007662	Paxs	433.22	Joback Method
dvisc	0.0005191	Paxs	477.67	Joback Method
dvisc	0.0003758	Paxs	522.12	Joback Method
dvisc	0.0002863	Paxs	566.58	Joback Method
dvisc	0.0002268	Paxs	611.03	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C18175218&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C18175218&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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